

42 Years in 42 Minutes

(more or less)

Carl Wamser
September 28, 2012

New York World's Fair



1939

Jelenko Laboratory



1940



1946 ?

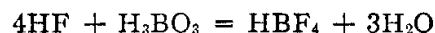
J. Amer. Chem. Soc., 70, 1209-1215 (1948)

[CONTRIBUTION FROM THE LABORATORY OF C. A. WAMSER]

Hydrolysis of Fluoboric Acid in Aqueous Solution

BY CHRISTIAN A. WAMSER

Aqueous fluoboric acid, which is now commercially available in the form of a 40% concentrate, is prepared¹ by treating aqueous hydrofluoric acid with the calculated amount of boric acid according to the equation



It has been well known that aqueous solutions of fluoboric acid are more or less strongly hydrolyzed depending on the dilution, temperature and length of time they have been standing after preparation. Such solutions always contain more H ion than can be accounted for by the fluoboric acid they are calculated to contain.

Any successful explanation of the net changes occurring on hydrolysis must be capable of accounting for the following experimentally observable facts:

(1) When four moles of hydrofluoric acid and one

(1) F. Fischer and K. Thiele, *Z. anorg. Chem.*, **67**, 304 (1910).

mole of boric acid are mixed in aqueous medium, an immediate sharp increase in conductance occurs and heat is evolved, yet the solution contains no fluoboric acid immediately after the preparation (no precipitate with nitron). The total titratable acidity (as measured by the volume of standard alkali consumed to the phenolphthalein end-point in the presence of mannitol) of such a mixture immediately after preparation corresponds to five equivalents of acid, the solution at the end-point containing only F^- and BO_2^- anions.

(2) After preparation, the fluoboric acid content gradually increases to a final (equilibrium) value, while the total acidity decreases to a definite value. The only anions present at the titration end-point are BF_4^- , F^- and BO_2^- .

(3) When a fluoboric acid concentrate (which is itself appreciably hydrolyzed) is diluted with water, the total acidity gradually increases to a final (equilibrium) value.



1978

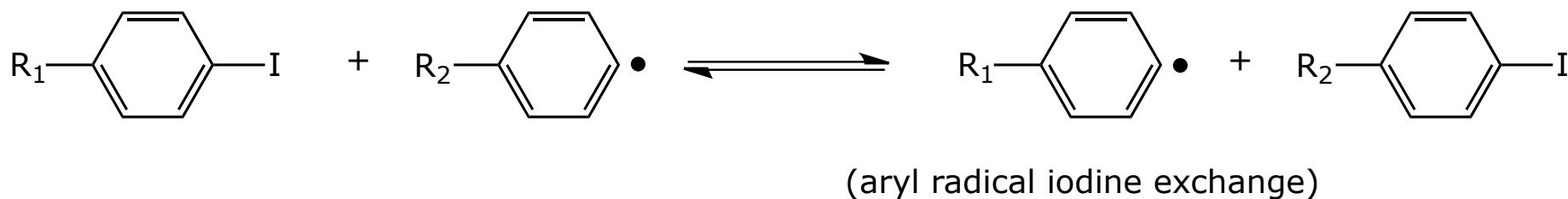
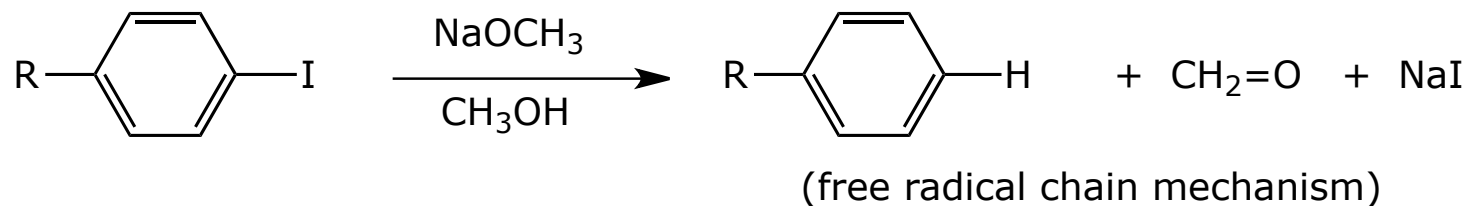
Brown University

Sc.B. 1966



Senior Honors Thesis

Joseph F. Bunnett



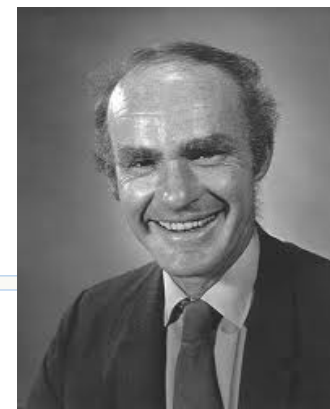
- **Radical Abstraction of Iodine from Aryl Iodides,**
J. F. Bunnett and C. C. Wamser, *J. Amer. Chem. Soc.* **1966**, 88, 5534-7.
- **Radical-Induced Deiodination of Aryl Iodides in Alkaline Methanol,**
J. F. Bunnett and C. C. Wamser, *J. Amer. Chem. Soc.* **1967**, 89, 6712-8.

1962-66

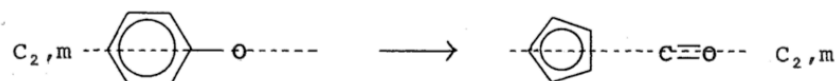
Caltech Ph.D. 1970



Caltech



George S. Hammond



<u>Phenoxy Radical</u>				<u>Products</u>			
		<u>C₂</u>	<u>m</u>			<u>C₂</u>	<u>m</u>
ψ ₁		+	+	φ ₁		+	+
ψ ₂		-	-	φ ₂		-	-
ψ ₃		-	+	φ ₃		-	+
ψ ₄		-	+	φ ₄		-	+
ψ ₅		+	-	φ ₅		+	-
ψ ₆		-	+	φ ₆		-	+
ψ ₇		+	-	φ ₇		+	-
ψ ₈		-	+	φ ₈		-	+
ψ ₉		-	+	φ ₉		-	+
ψ ₁₀		+	+	φ ₁₀		-	-
ψ ₁₁		-	-	φ ₁₁		-	-

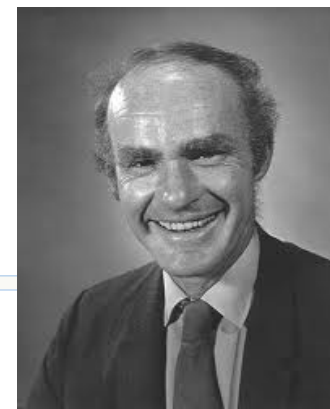
Thesis Project 1

1966-69

Caltech
Ph.D. 1970

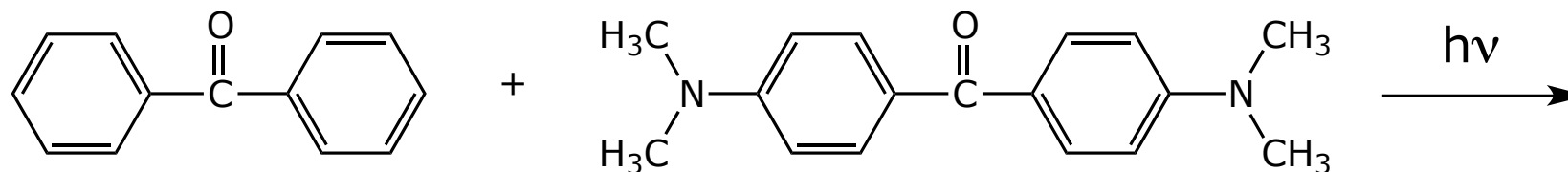


Caltech



Thesis Project 2

George S. Hammond

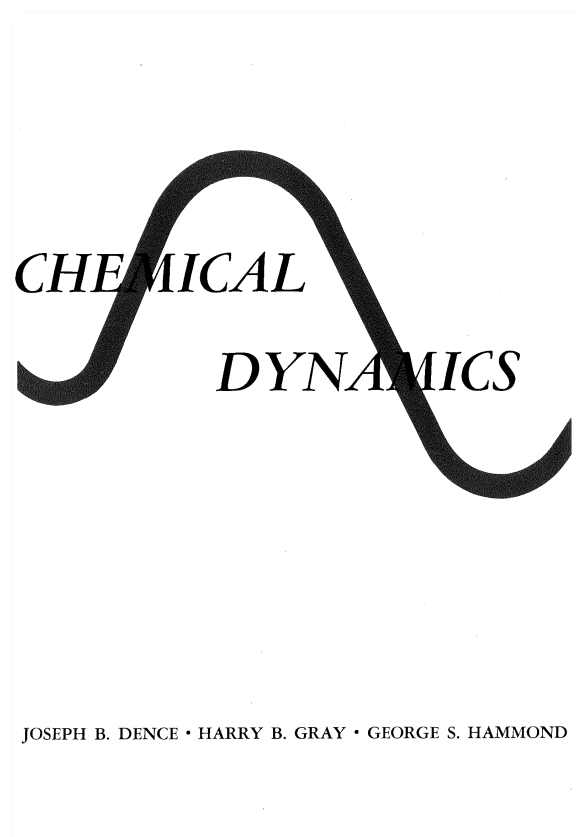


- **The Photoreaction of Michler's Ketone with Benzophenone - A Triplet Exciplex,**
C. C. Wamser, G. S. Hammond, C. T. Chang, and C. Baylor, Jr.,
J. Amer. Chem. Soc. **1970**, 92, 6362-3.

1966-69

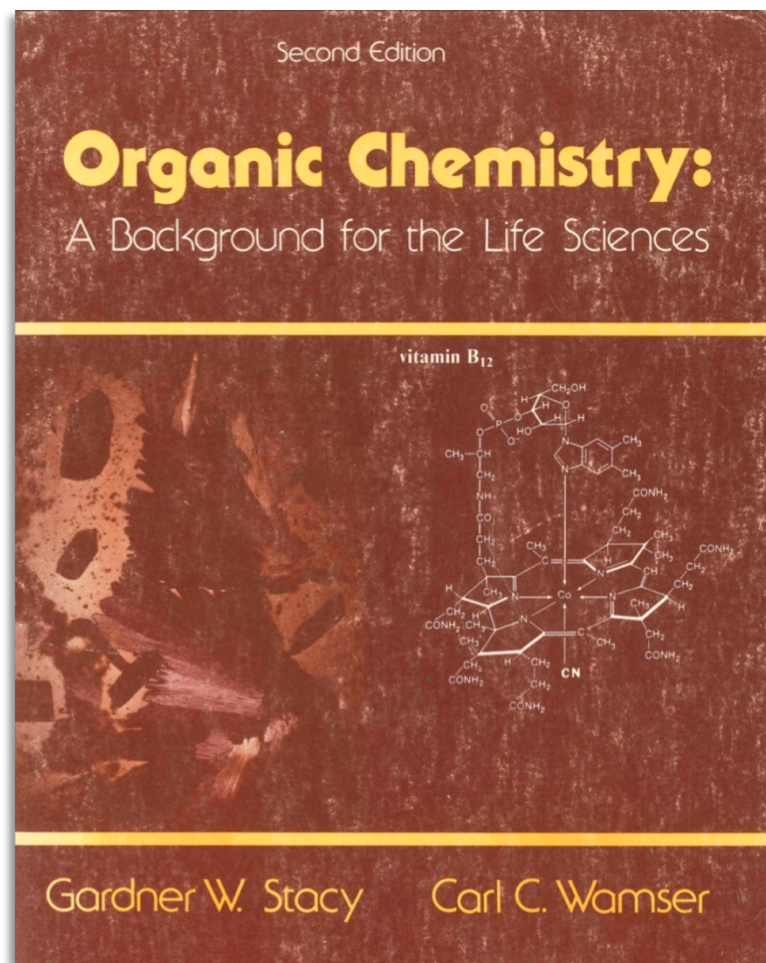
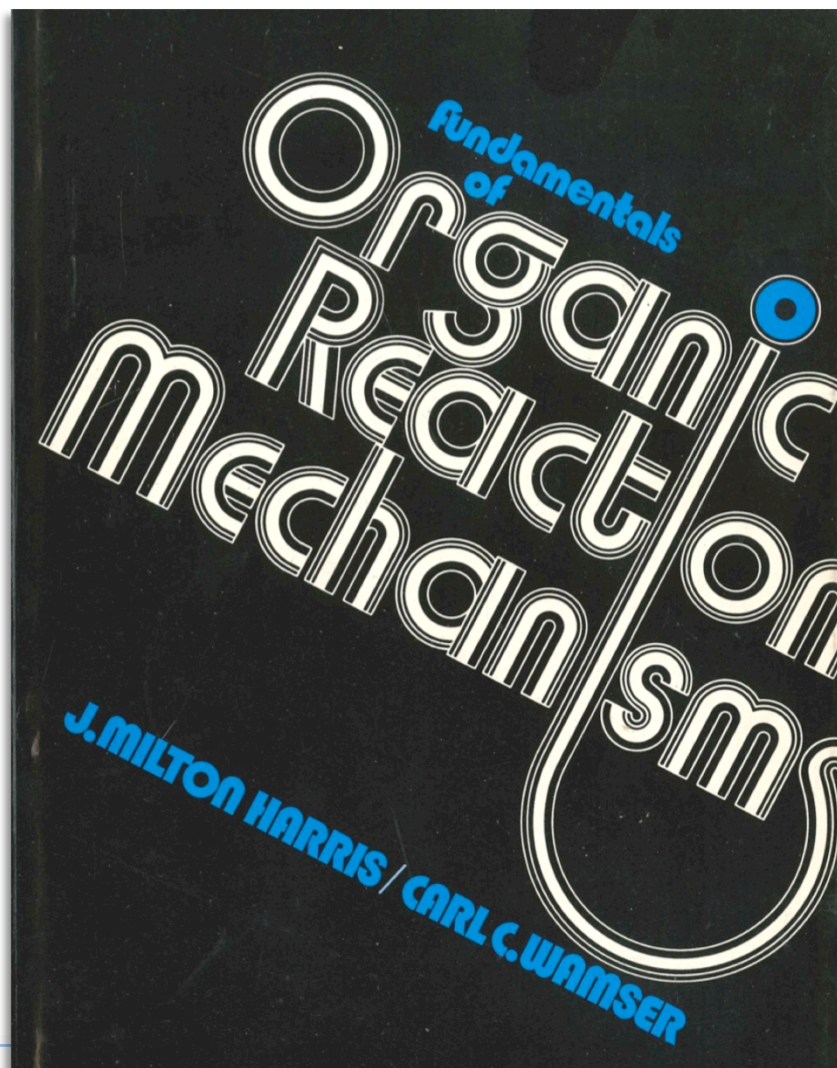
Caltech

The Hammond-Gray Curriculum



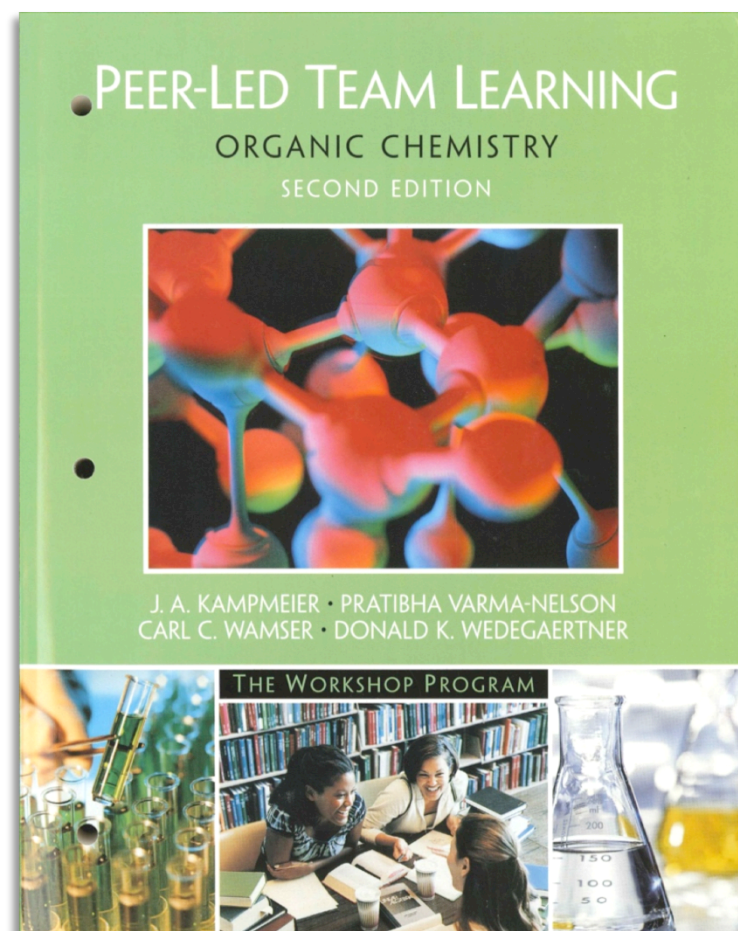
1966-69

Textbook Writing Projects



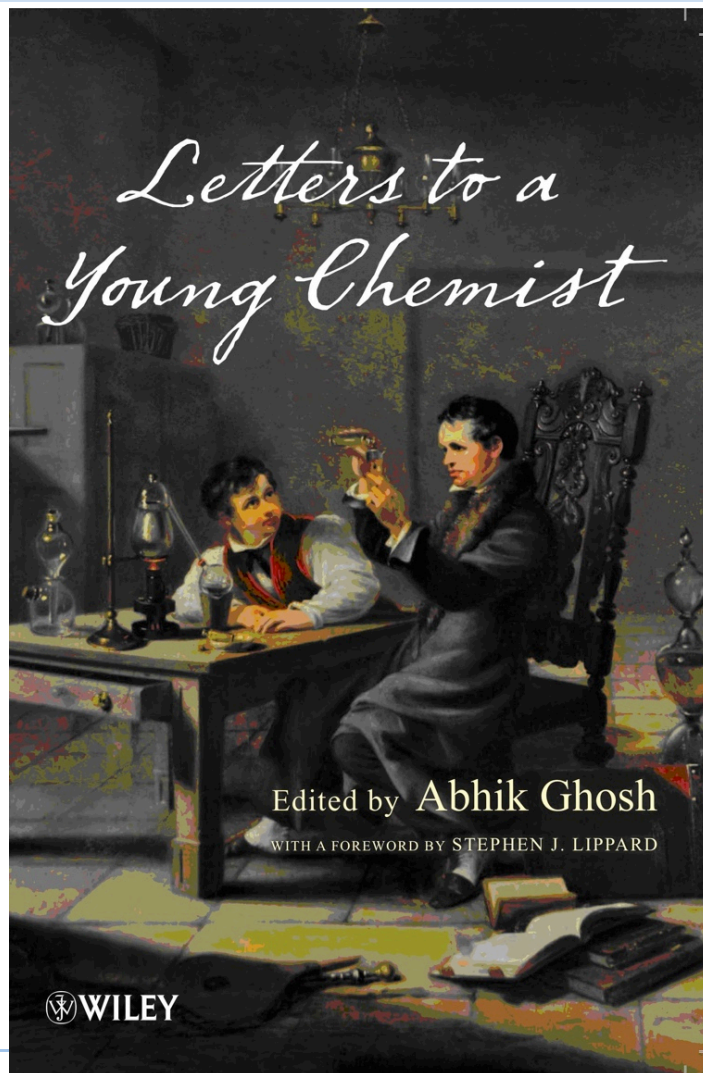
1976, 1985

Workbook Writing Projects



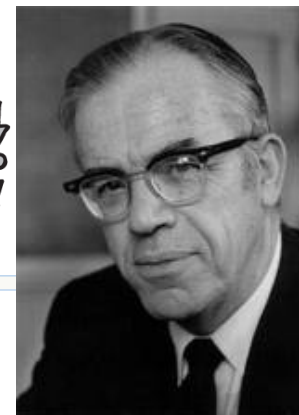
2001, 2006

Latest Writing Project



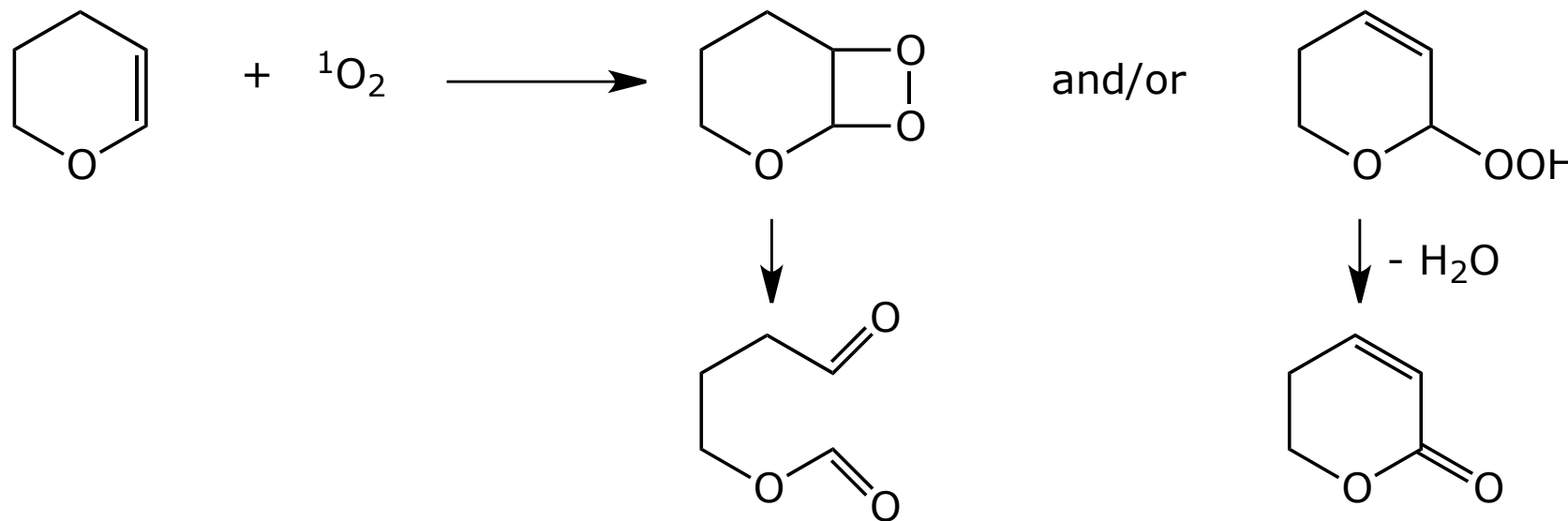
2011

Harvard University Postdoctoral Research



Paul D. Bartlett

Singlet Oxygen Chemistry



Reaction of Singlet Oxygen with 4-Methyl-2,3-dihydro-g-pyrans, A. A. Frimer,
P. D. Bartlett, A. F. Boschung, J. G. Jewett, *J. Amer. Chem. Soc.*, **1977**, 99, 7977-7986.

1969-70

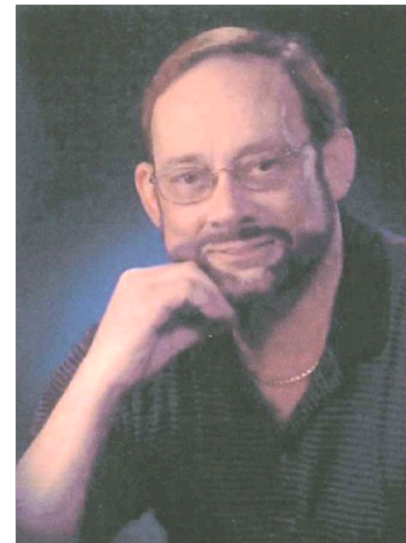
California State University, Fullerton



Andy Montana
Department Chair



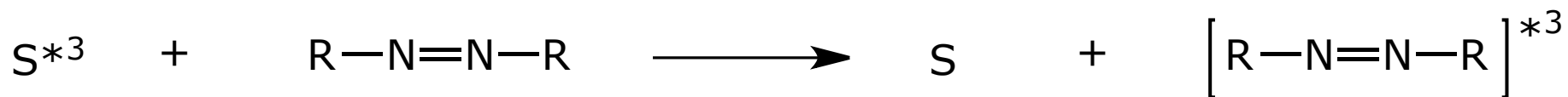
CALIFORNIA STATE UNIVERSITY
FULLERTON



Mike King
First Research Student

The First Papers – Energy Transfer

Flash Photolysis

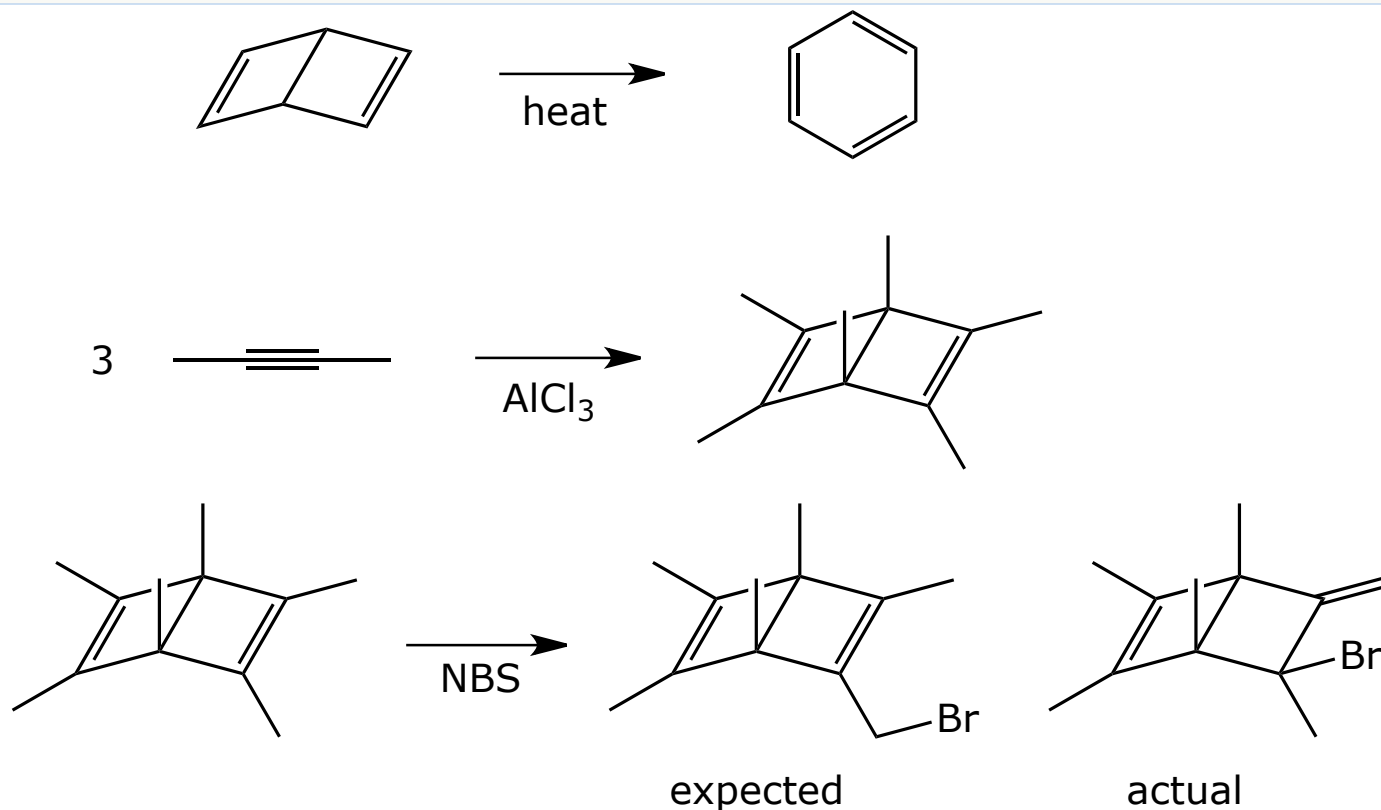


R = n-Bu, i-Bu, s-Bu, t-Bu

- **Steric Effects in Singlet and Triplet Electronic Energy Transfer,**
C. C. Wamser and P. L. Chang, *J. Amer. Chem. Soc.* **1973**, 95, 2044-5.
- **Steric Effects in Singlet and Triplet Electronic Energy Transfer to Azo Compounds,**
C. C. Wamser, R. T. Medary, I. E. Kochevar, N. J. Turro, and P. L. Chang,
J. Amer. Chem. Soc. **1975**, 97, 4864-9.
- **Singlet Electronic Energy Transfer to Azoalkanes; Separation of Collisional and Long-Range Mechanisms by Steric and Solvent Viscosity Effects,** C. C. Wamser, L. Lou, J. Mendoza, and E. Olson, *J. Amer. Chem. Soc.* **1981**, 103, 7228-32.

1970s

Dewar Benzenes



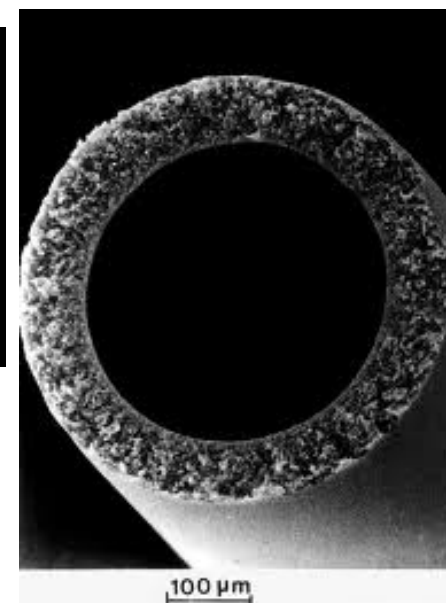
- **Hexamethyl(Dewar Benzene)**, S. A. Shama and C. C. Wamser, *Org. Synth.* **1983**, 61, 62-4.
- **Synthesis and Reactions of 5-Methylenebicyclo[2.2.0]hex-2-ene Derivatives from Hexamethyl(Dewar Benzene)**, C. C. Wamser, D. D. Ngo, M. J. Rodriguez, S. A. Shama, and T. L. Tran, *J. Amer. Chem. Soc.* **1989**, 111, 2162-2168.

Sabbatical, UC Berkeley



Melvin Calvin

Artificial Photosynthesis



- **Preparation and Properties of Porphyrin-Modified Hollow Fiber Membranes as Photosensitizers for Singlet Oxygen and for Artificial Photosynthesis,**
C. C. Wamser, M. Calvin, and G. Graf, *J. Membr. Sci.* **1986**, 28, 31-46.

1980

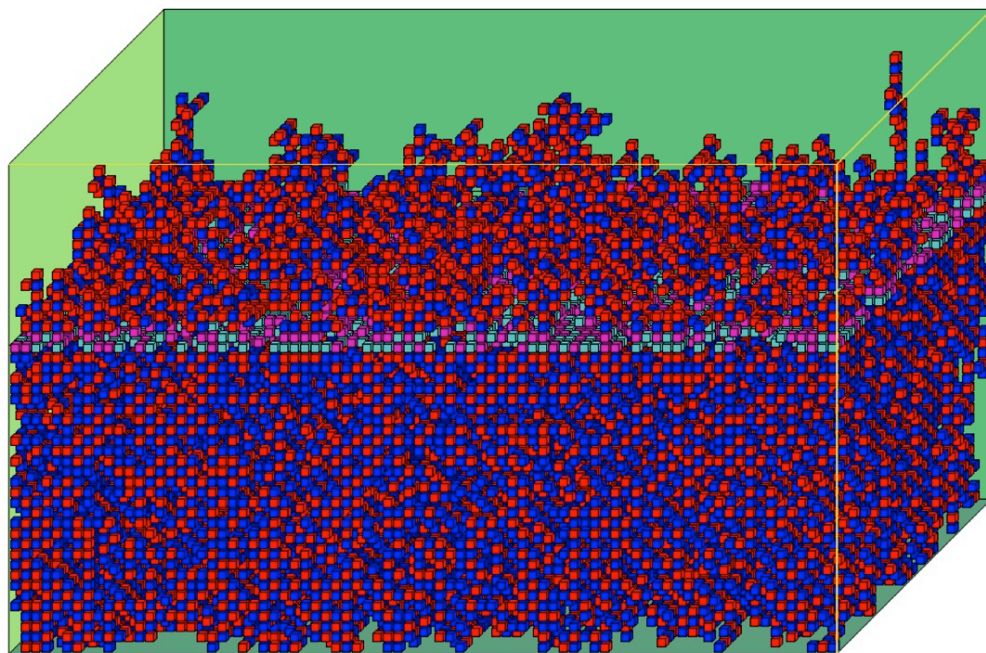
Portland State University



Portland State University

Interfacial Polymerization

Bend Research, Inc.



- TAPP in H₂O (pH 3)

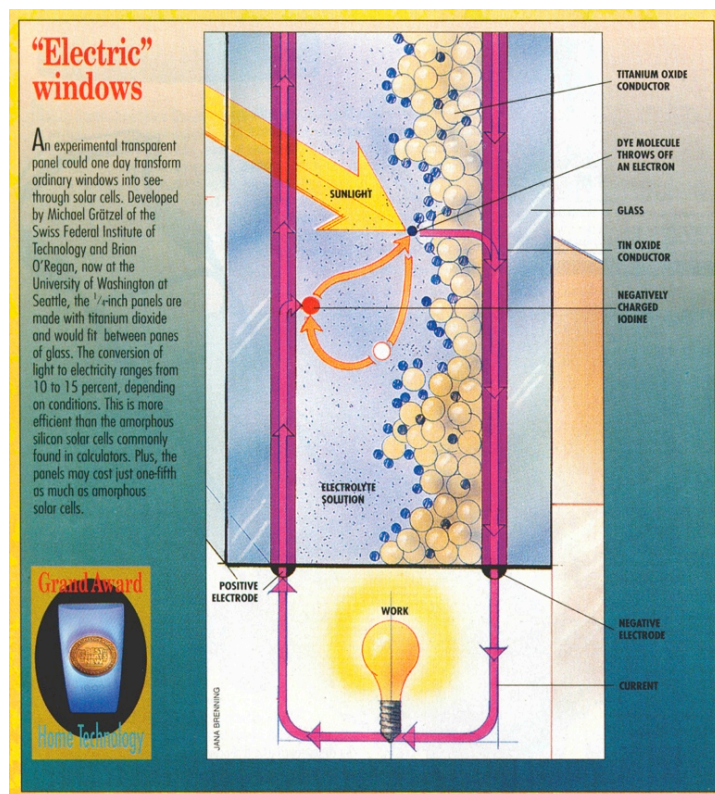
- TCCPP in CH₂Cl₂

- **Synthesis and Photoactivity of Chemically Asymmetric Polymeric Porphyrin Films Made by Interfacial Polymerization**, C. C. Wamser, R. R. Bard, V. Senthilathipan, V. C. Anderson, J. A. Yates, H. K. Lonsdale, G. W. Rayfield, D. T. Friesen, D. A. Lorenz, G. C. Stangle, P. van Eikeren, D. R. Baer, R. A. Ransdell, J. H. Golbeck, W. C. Babcock, J. J. Sandberg, and S. E. Clarke, *J. Amer. Chem. Soc.* **1989**, 111, 8485-8492. 1980s

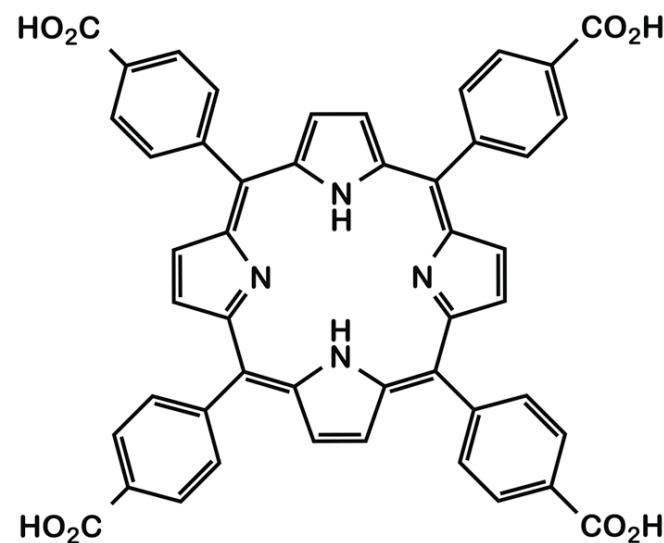
Sabbatical, EPFL



Dye-Sensitized Solar Cells

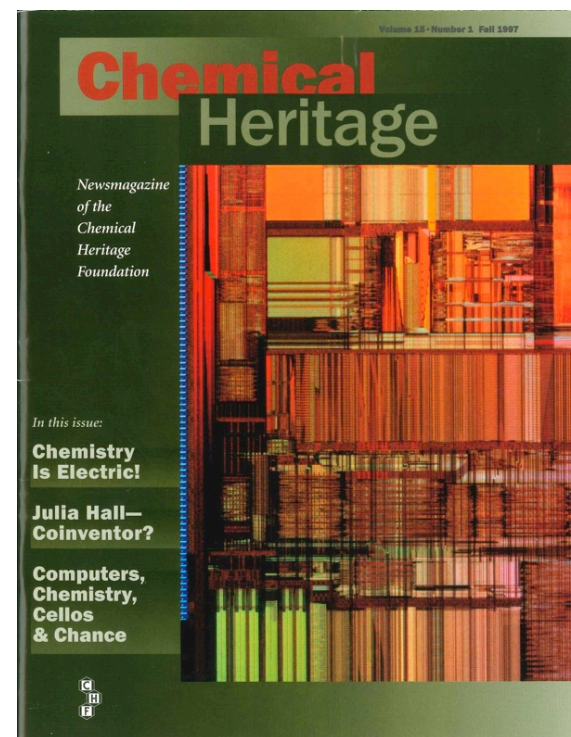
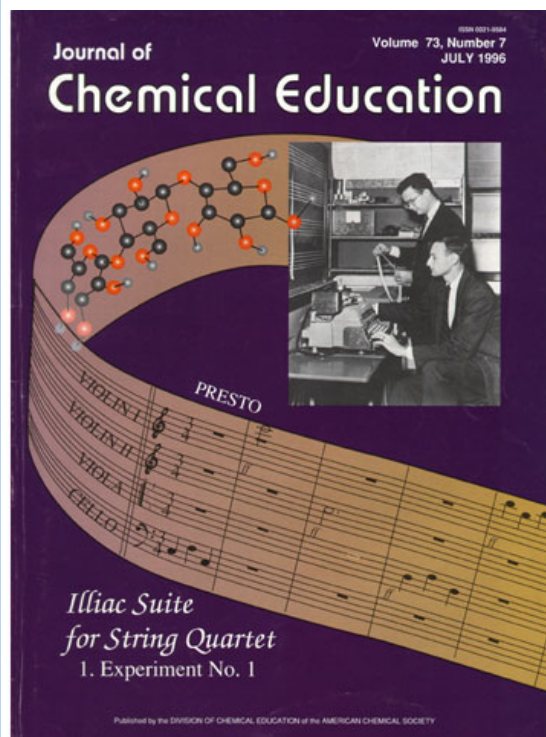


Michael Grätzel



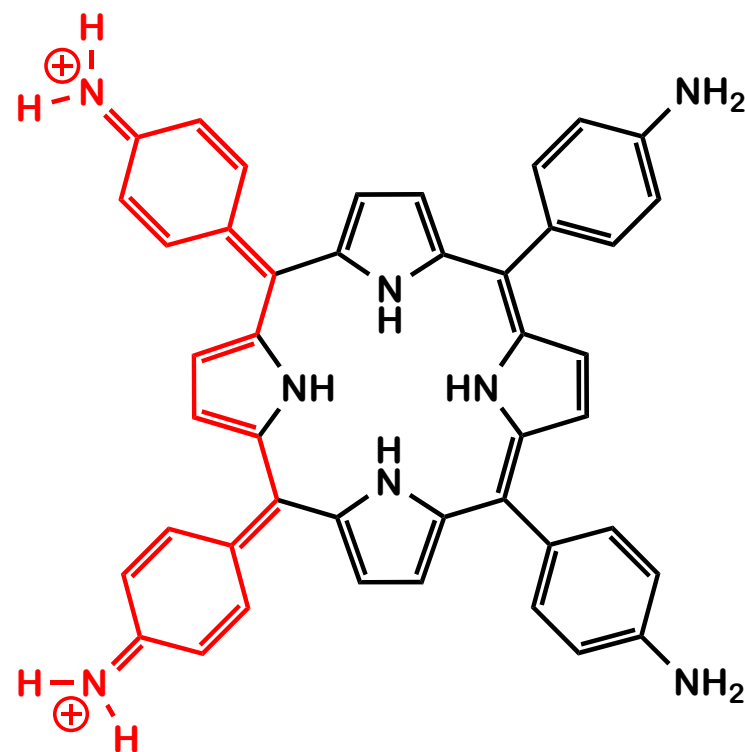
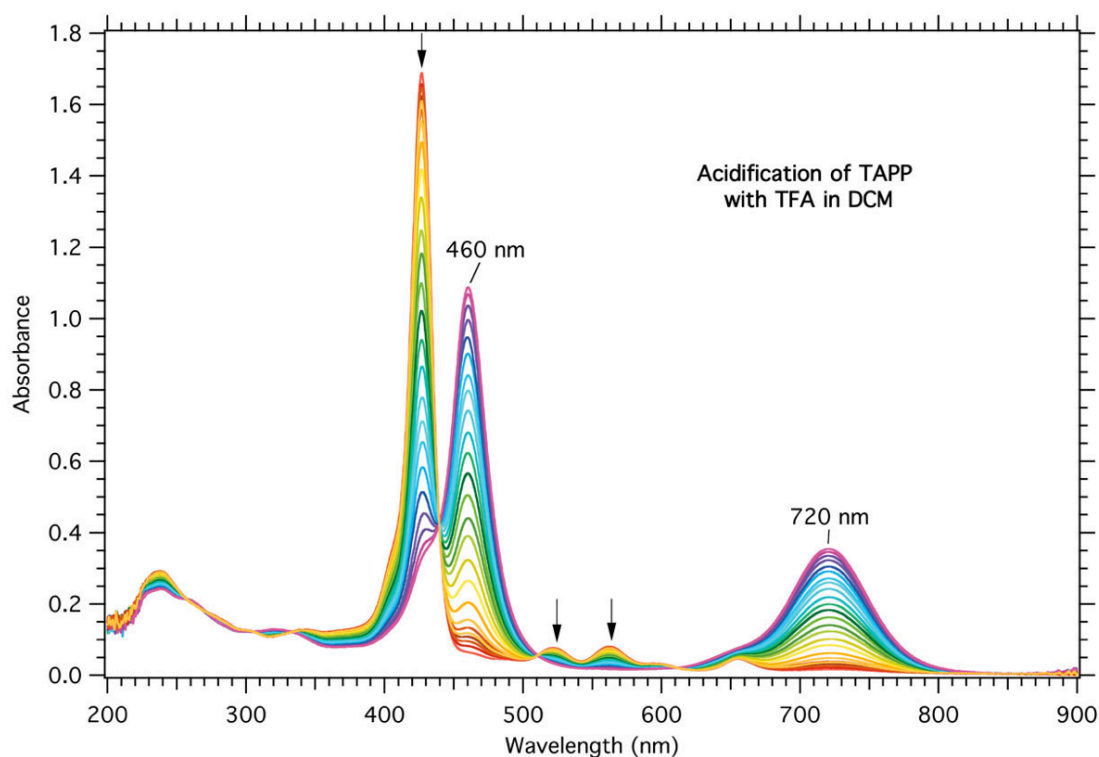
Adsorption and Photoactivity of Tetra(4-carboxyphenyl)porphyrin (TCPP) on Nano-particulate TiO_2 , S. Cherian and C. C. Wamser, *J. Phys. Chem. B*, **2000**, 104(15), 3624-9. 1992

Chemistry and Music: Collaborations with My Father



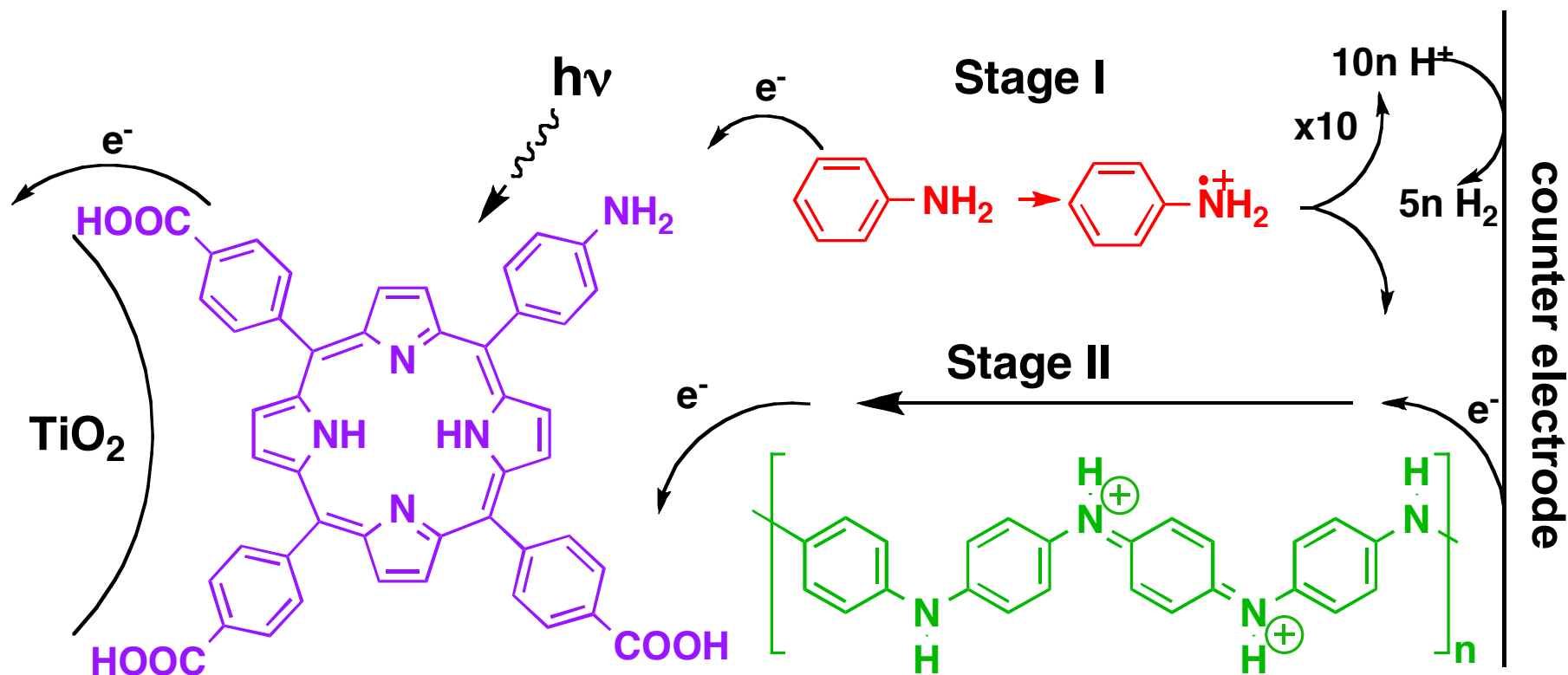
- **Lejaren A. Hiller, Jr.: A Memorial Tribute to a Modern Chemist-Composer**
C. A. Wamser and C. C. Wamser, *J. Chem. Educ.* **1996**, 73(7), 601-607.
- **Computers, Chemistry, Cellos, and Chance: A Celebration of Lejaren A. Hiller, Jr.**
C. C. Wamser and C. A. Wamser, *Chem. Heritage*, **1997**, 15(1), 8-9, 34-35.

Hyperporphyrin Spectroscopy



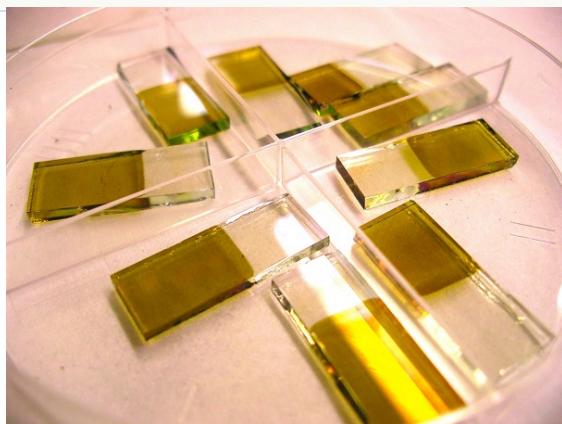
- **Substituent and Solvent Effects on the Hyperporphyrin Spectra of Diprotonated Tetraphenylporphyrins**, J. Weinkauff, A. Schweiger, S. Cooper, and C. C. Wamser, *J. Phys. Chem. A*, **2003**, 107 (18), 3486-3496.

Solid-State Grätzel Cells

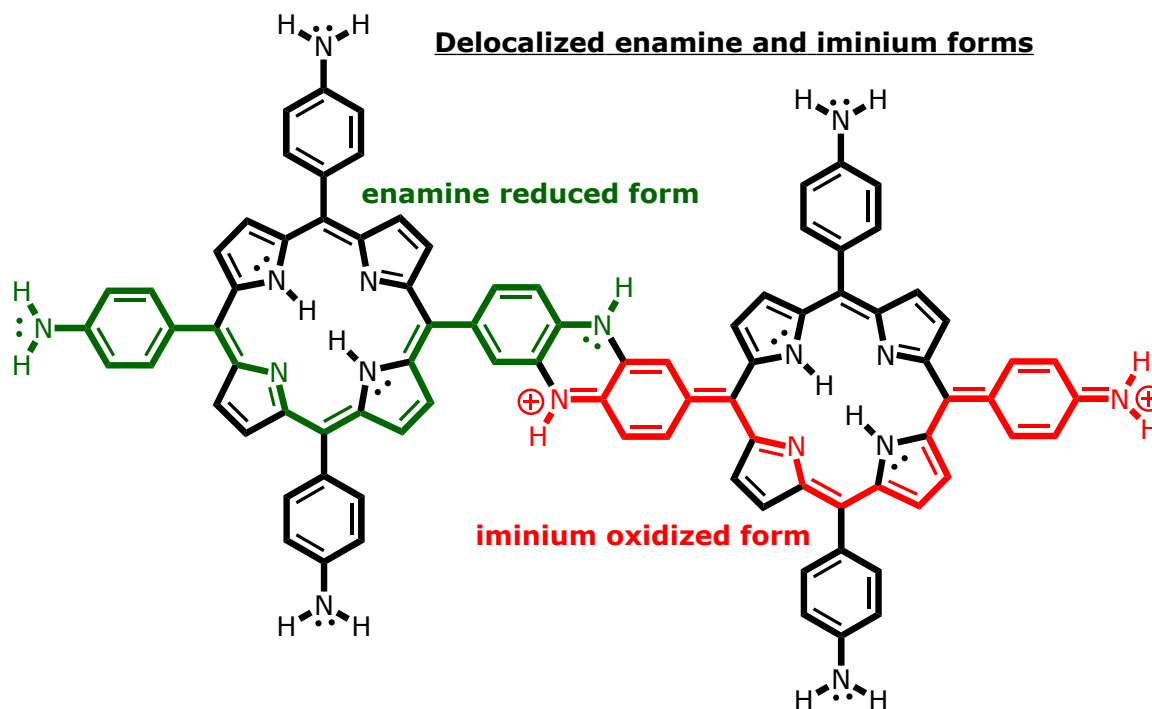
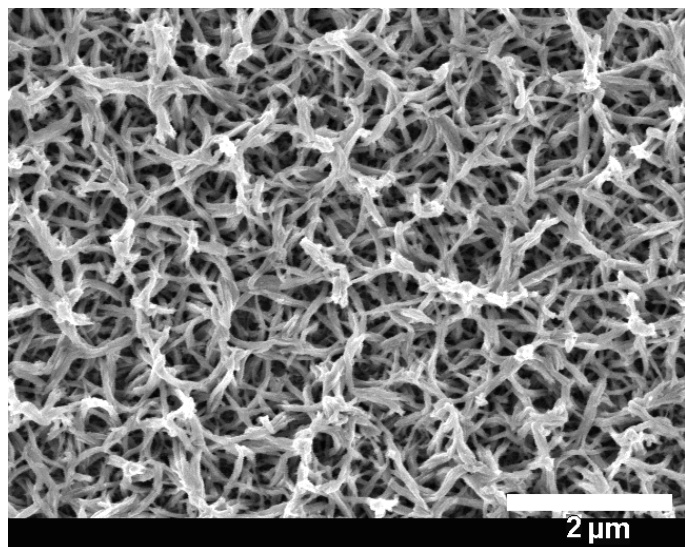


- **Photoelectropolymerization of Aniline in a Dye-Sensitized Solar Cell,**
H.-S. Kim and C. C. Wamser, *Photochem. Photobiol. Sci.*, **2006**, 5 (10), 955-960.

Inverse Grätzel Cells



Poly-TAPP



Synthesis and Characterization of Electropolymerized Nanostructured Aminophenylporphyrin Films, M. G. Walter and C. C. Wamser, *J. Phys. Chem. C*, **2010**, 114, 7563-7574.

2010

The Journal of Organic Chemistry

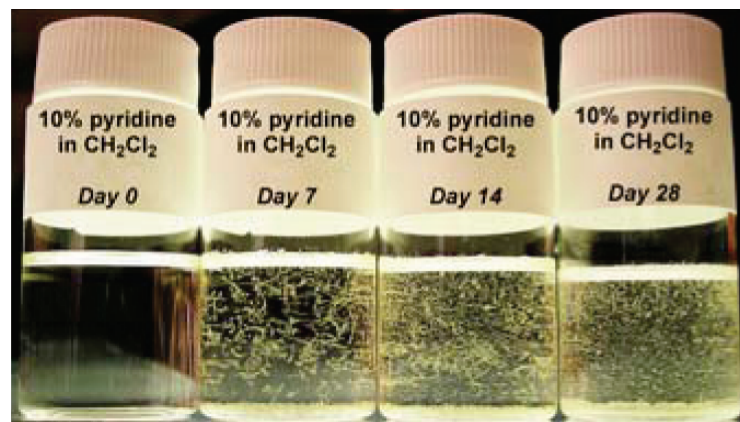
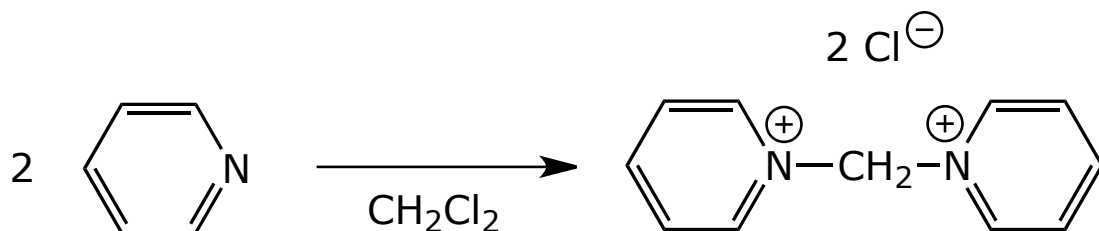
Listed below are the most accessed *JOC* articles during the second quarter of 2010:

Reaction of Dichloromethane with Pyridine Derivatives under Ambient Conditions

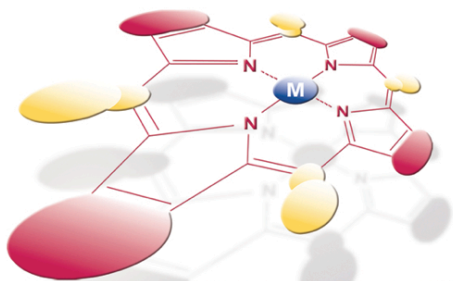
Alexander B. Rudine, Michael G. Walter, and Carl C. Wamser

J. Org. Chem., **2010**, 75 (12), pp 4292–4295

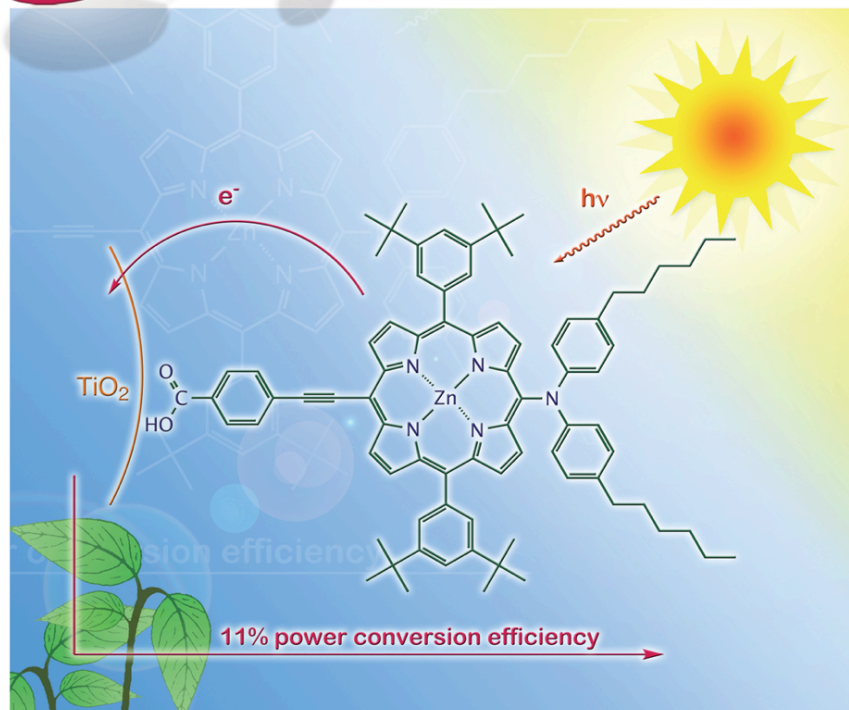
DOI: 10.1021/jo100276m



The official journal of SPP
**Journal of
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chemistry, physics, biology
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Society of
Porphyrins &
Phthalocyanines

Review Article

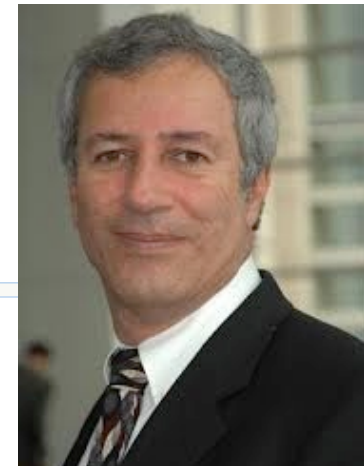
Porphyrins and Phthalocyanines in Solar Photovoltaic Cells

Alexander B. Rudine,
Michael G. Walter, and
Carl C. Wamser

J. Porphyrins Phthalocyanines,
2010; 14: 759–792.

Sabbatical Caltech

Polymer Solar Cells



Nate Lewis

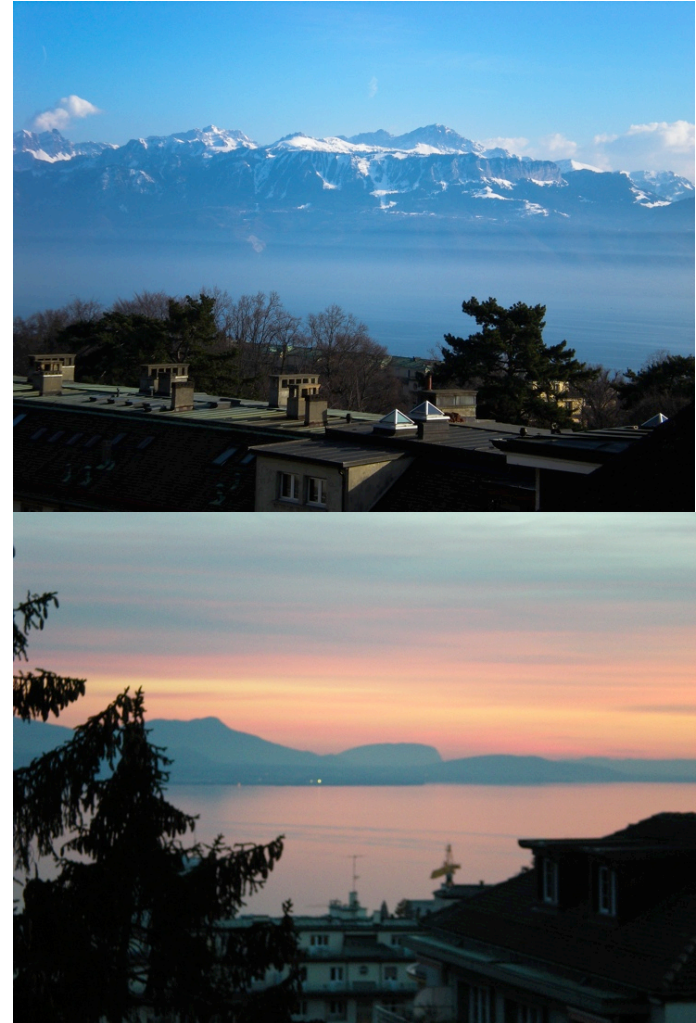




Moscow, 2008

Sabbatical
EPFL+

Lausanne, Switzerland



2011

Sabbatical EPFL+

University of Nantes, France



Fabrice Odobel



2011

Sabbatical EPFL+

Valencia, Spain



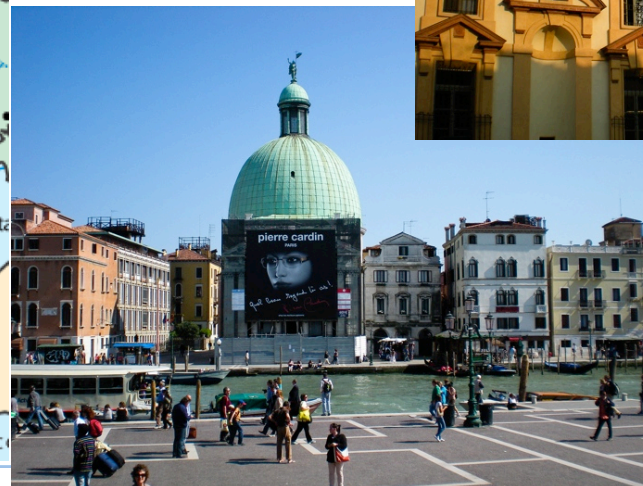
3rd HYBRID AND ORGANIC PHOTOVOLTAICS CONFERENCE
HOPV2011
15 - 18 MAY 2011 - VALENCIA - SPAIN



2011

Sabbatical EPFL+

Italy - Milan, Pavia, Venice



2011

Sabbatical EPFL+

University of Tromsø, Norway



Abhik Ghosh



2011

Sabbatical EPFL+

University of Edinburgh, Scotland



2011

PSU Photovoltaic Test Facility



Co-PIs:

David Sailor

Todd Rosenstiel
(Erik Johansson)

Sponsors:

NSF

PGE

OR BEST

PSU ISS

Portland BES

PSU OGSR

EC Company

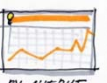
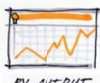
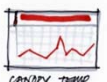
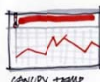
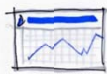
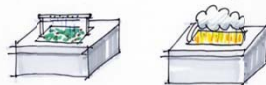
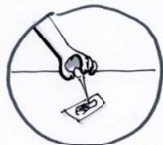
UO SRML

NSF - CRPA

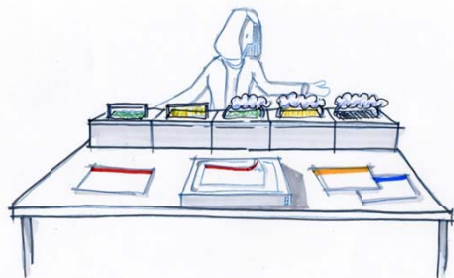


Connecting Researchers to Public Audiences

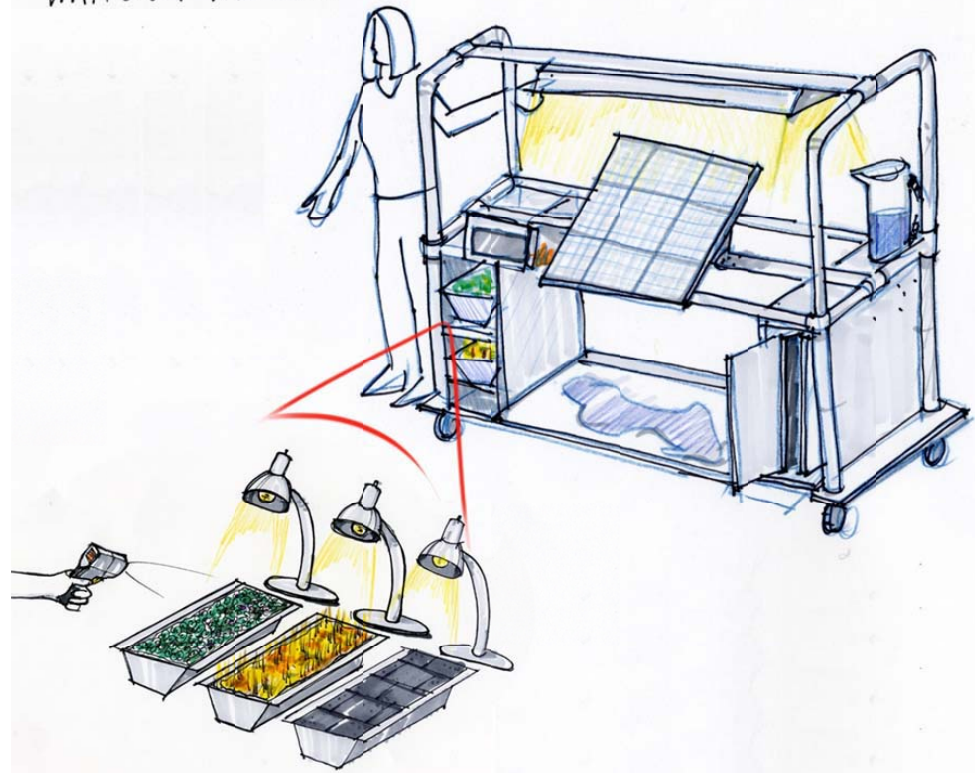
WHAT DYE IS BEST?



DATA ANALYSIS of GREEN ROOF VARIABLES



WATT'S ON THE ROOF



Humanity's Top Ten Problems for the next 50 years

1. ENERGY
2. WATER
3. FOOD
4. ENVIRONMENT
5. POVERTY
6. TERRORISM & WAR
7. DISEASE
8. EDUCATION
9. DEMOCRACY
10. POPULATION



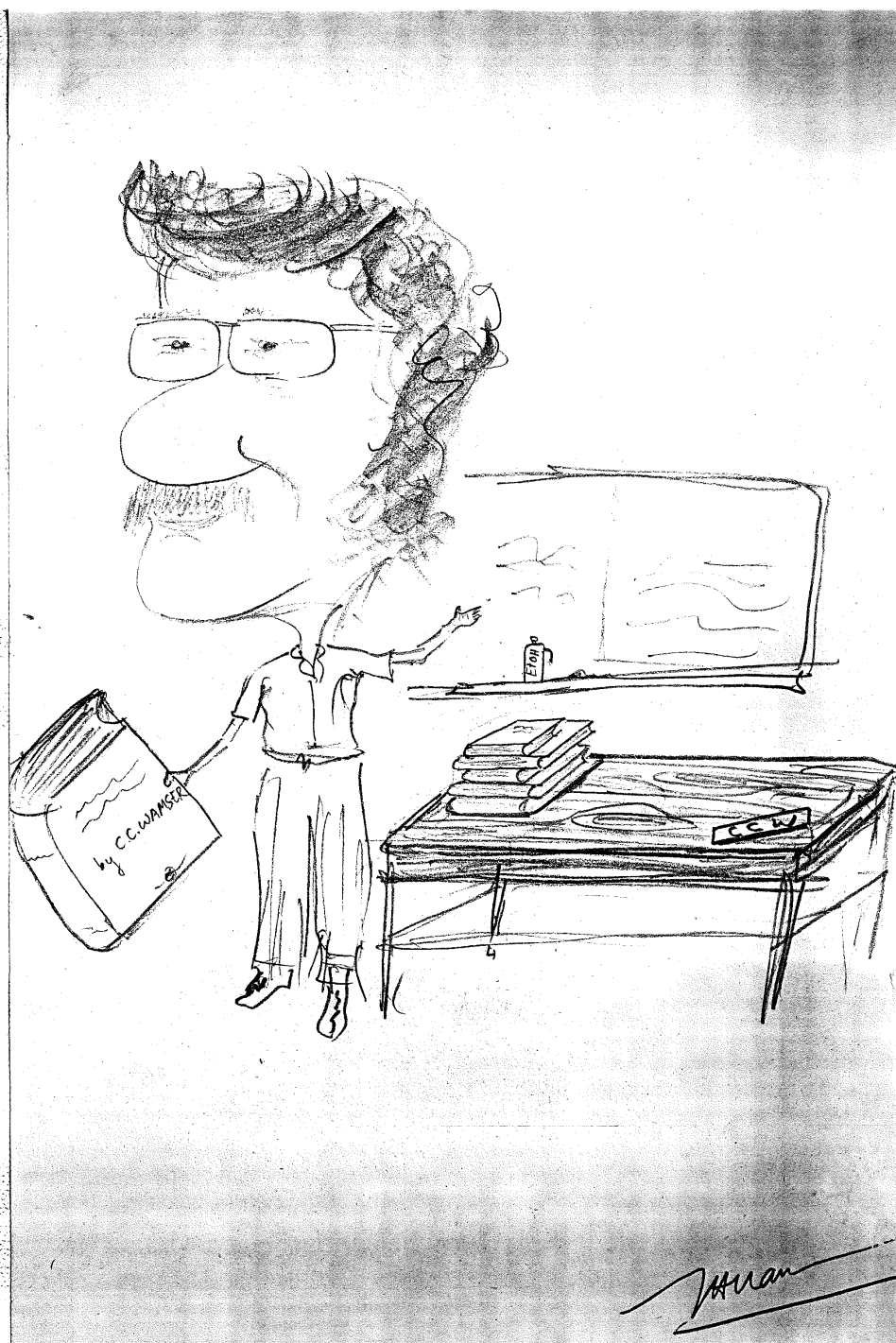
Partners in Science
January 18, 2003

R. E. Smalley
Rice University

O Grignard

The carbonyl is polarized
The carbon end is plus.
The nucleophile will thus attack
The carbon nucleus.
The Grignard makes an alcohol,
Of types there are but three.
It makes a bond
To correspond
From C to shining C.

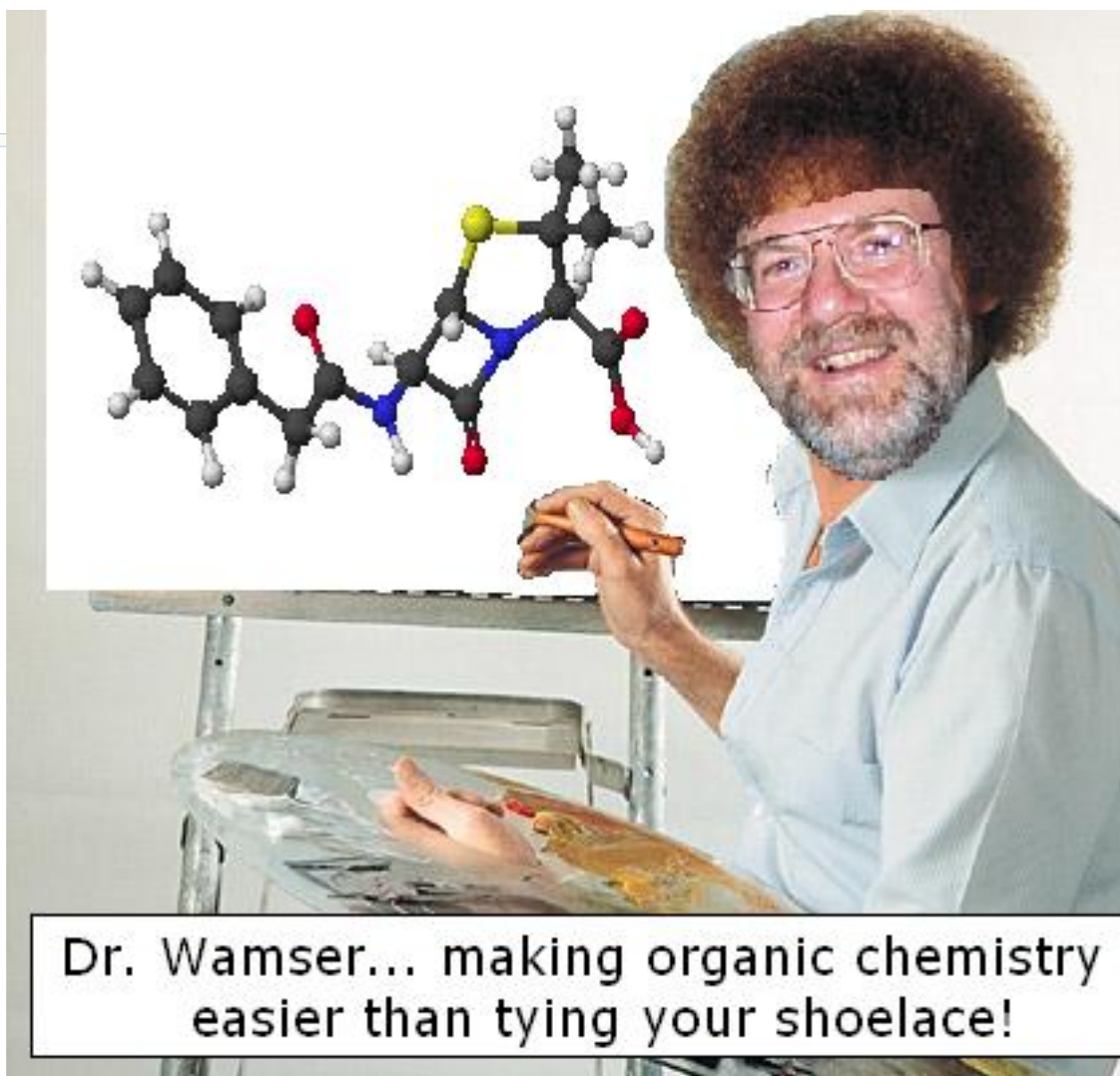
Lyrics by Frank Westheimer (Harvard Univ)





CARL OF THE (CARBON) RINGS

THE MESO TOWERS



Organic Chemistry – Spring 2012





