

# 42 Years in 42 Minutes

(more or less)

*Carl Wamser*  
*September 28, 2012*

# New York World's Fair



1939

# Jelenko Laboratory



1940



1946 ?

[CONTRIBUTION FROM THE LABORATORY OF C. A. WAMSER]

## Hydrolysis of Fluoboric Acid in Aqueous Solution

BY CHRISTIAN A. WAMSER

Aqueous fluoboric acid, which is now commercially available in the form of a 40% concentrate, is prepared<sup>1</sup> by treating aqueous hydrofluoric acid with the calculated amount of boric acid according to the equation



It has been well known that aqueous solutions of fluoboric acid are more or less strongly hydrolyzed depending on the dilution, temperature and length of time they have been standing after preparation. Such solutions always contain more H ion than can be accounted for by the fluoboric acid they are calculated to contain.

Any successful explanation of the net changes occurring on hydrolysis must be capable of accounting for the following experimentally observable facts:

(1) When four moles of hydrofluoric acid and one

(1) F. Fischer and K. Thiele, *Z. anorg. Chem.*, **67**, 304 (1910).

mole of boric acid are mixed in aqueous medium, an immediate sharp increase in conductance occurs and heat is evolved, yet the solution contains no fluoboric acid immediately after the preparation (no precipitate with nitron). The total titratable acidity (as measured by the volume of standard alkali consumed to the phenolphthalein end-point in the presence of mannitol) of such a mixture immediately after preparation corresponds to five equivalents of acid, the solution at the end-point containing only F<sup>-</sup> and BO<sub>2</sub><sup>-</sup> anions.

(2) After preparation, the fluoboric acid content gradually increases to a final (equilibrium) value, while the total acidity decreases to a definite value. The only anions present at the titration end-point are BF<sub>4</sub><sup>-</sup>, F<sup>-</sup> and BO<sub>2</sub><sup>-</sup>.

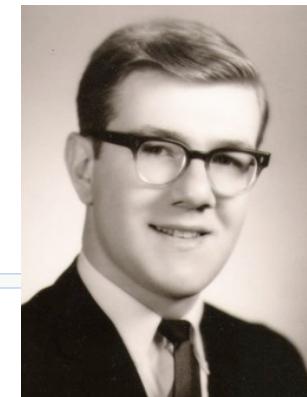
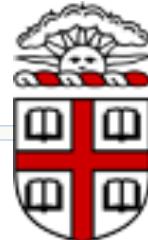
(3) When a fluoboric acid concentrate (which is itself appreciably hydrolyzed) is diluted with water, the total acidity gradually increases to a final (equilibrium) value.



1978

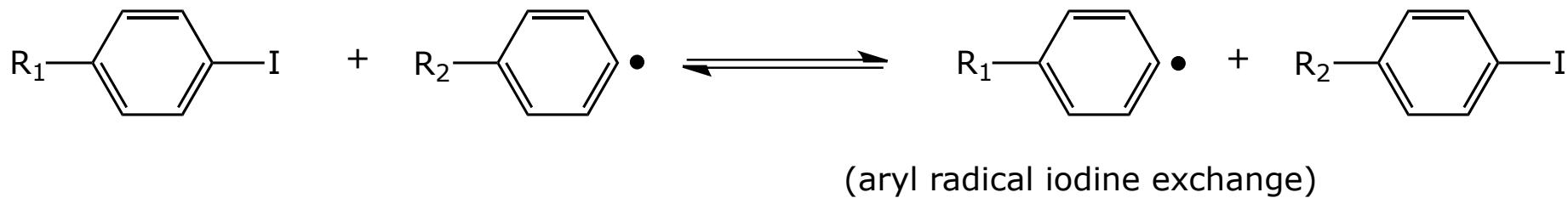
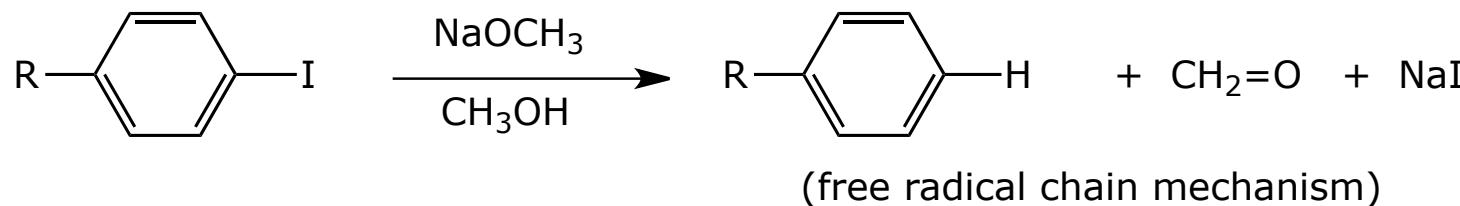
# Brown University

## Sc.B. 1966



Senior Honors Thesis

Joseph F. Bunnett



- **Radical Abstraction of Iodine from Aryl Iodides,**  
J. F. Bunnett and C. C. Wamser, *J. Amer. Chem. Soc.* **1966**, 88, 5534-7.
- **Radical-Induced Deiodination of Aryl Iodides in Alkaline Methanol,**  
J. F. Bunnett and C. C. Wamser, *J. Amer. Chem. Soc.* **1967**, 89, 6712-8.

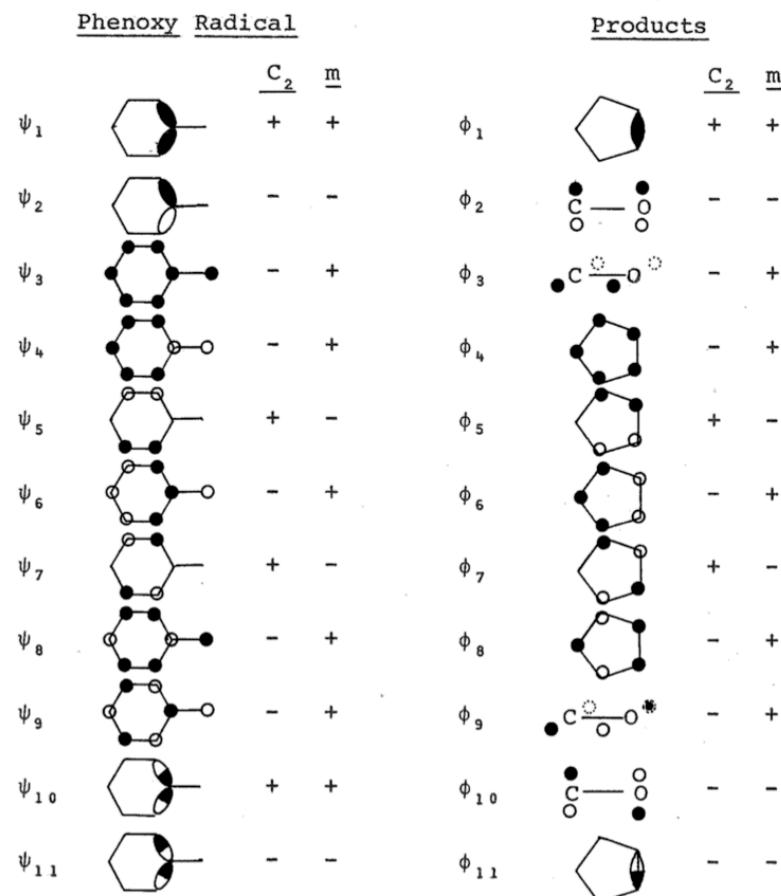
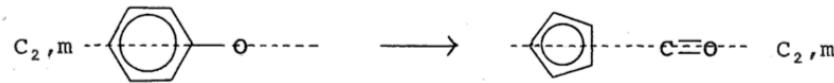
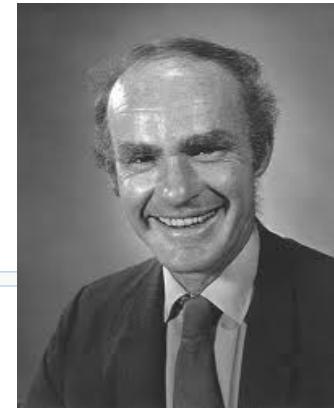
1962-66

# Caltech

## Ph.D. 1970



Caltech



George S. Hammond

Thesis Project 1

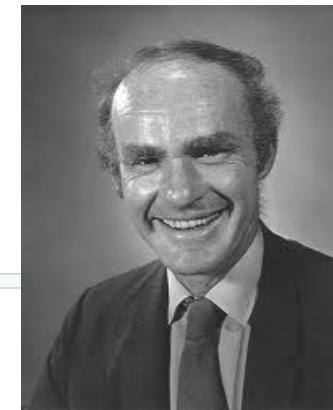
1966-69

# Caltech

## Ph.D. 1970

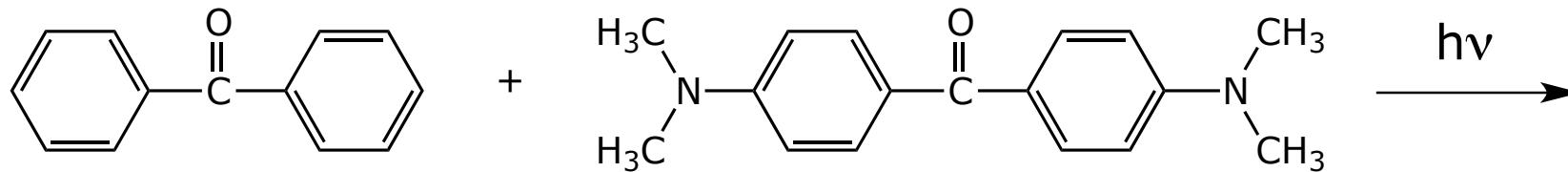


Caltech



Thesis Project 2

George S. Hammond

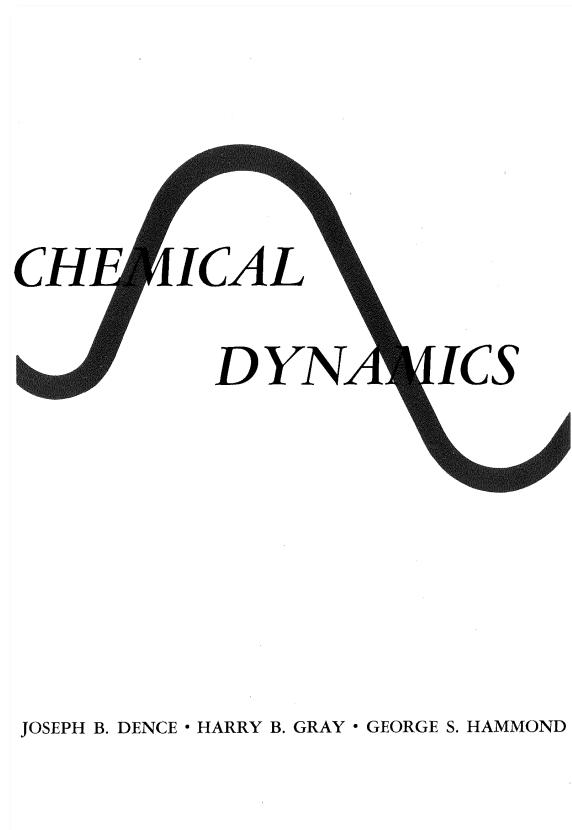


- **The Photoreaction of Michler's Ketone with Benzophenone - A Triplet Exciplex,**  
C. C. Wamser, G. S. Hammond, C. T. Chang, and C. Baylor, Jr.,  
*J. Amer. Chem. Soc.* **1970**, 92, 6362-3.

1966-69

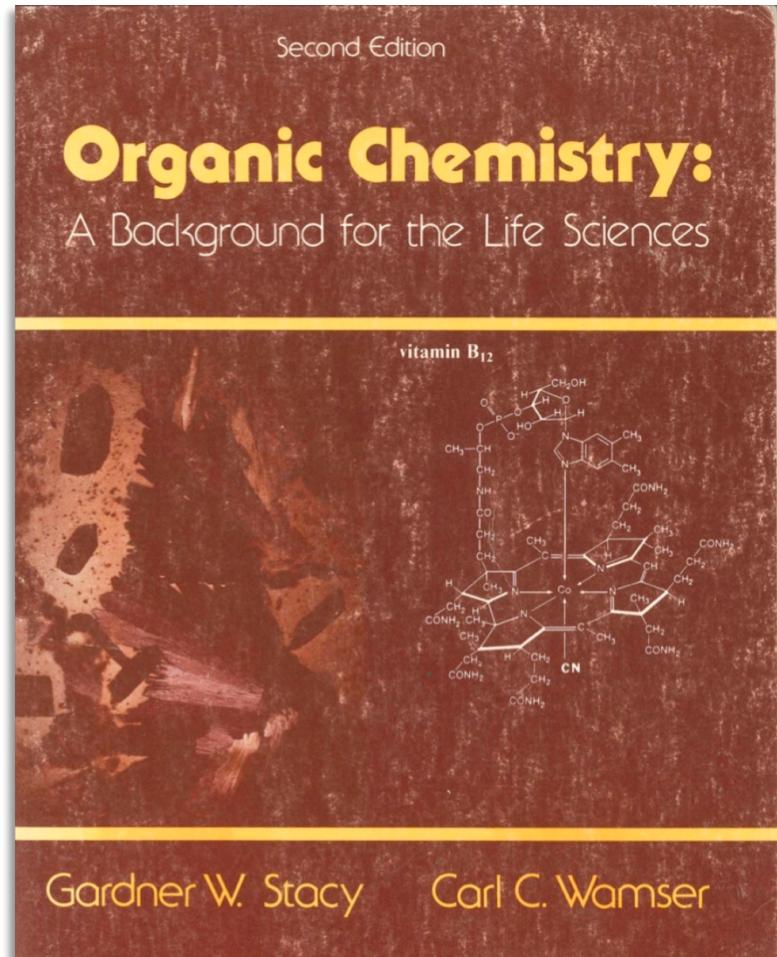
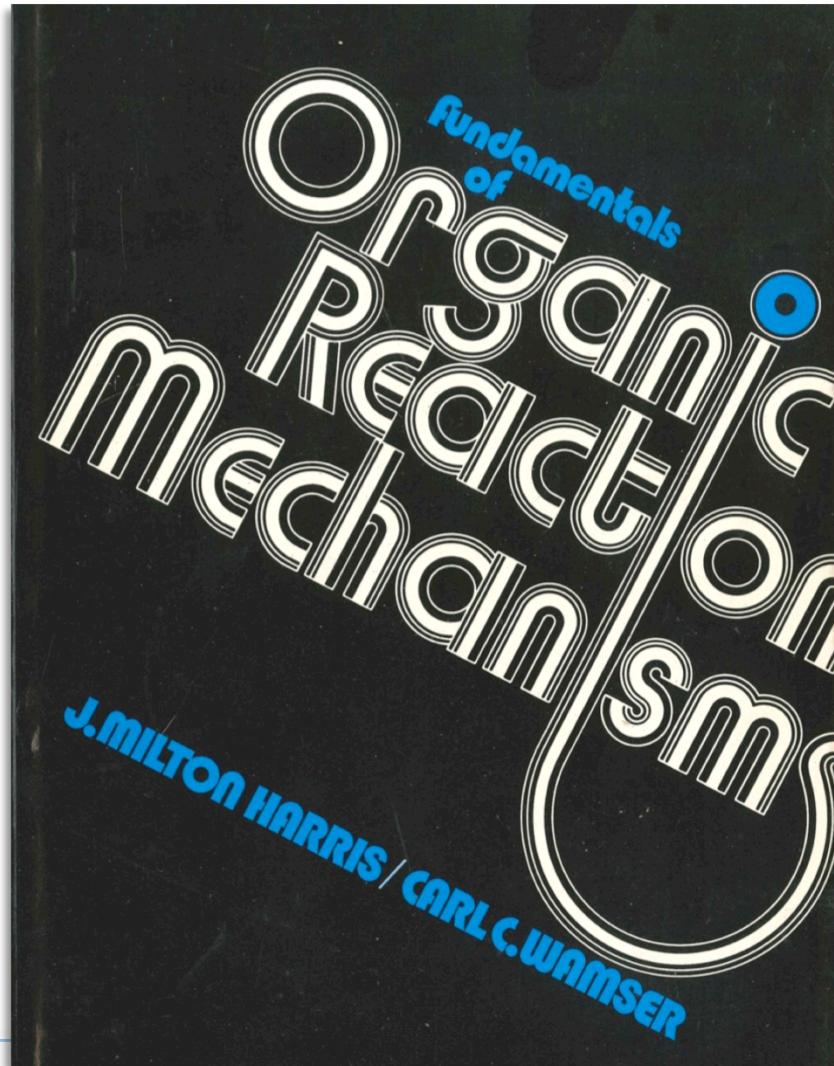
# Caltech

## The Hammond-Gray Curriculum



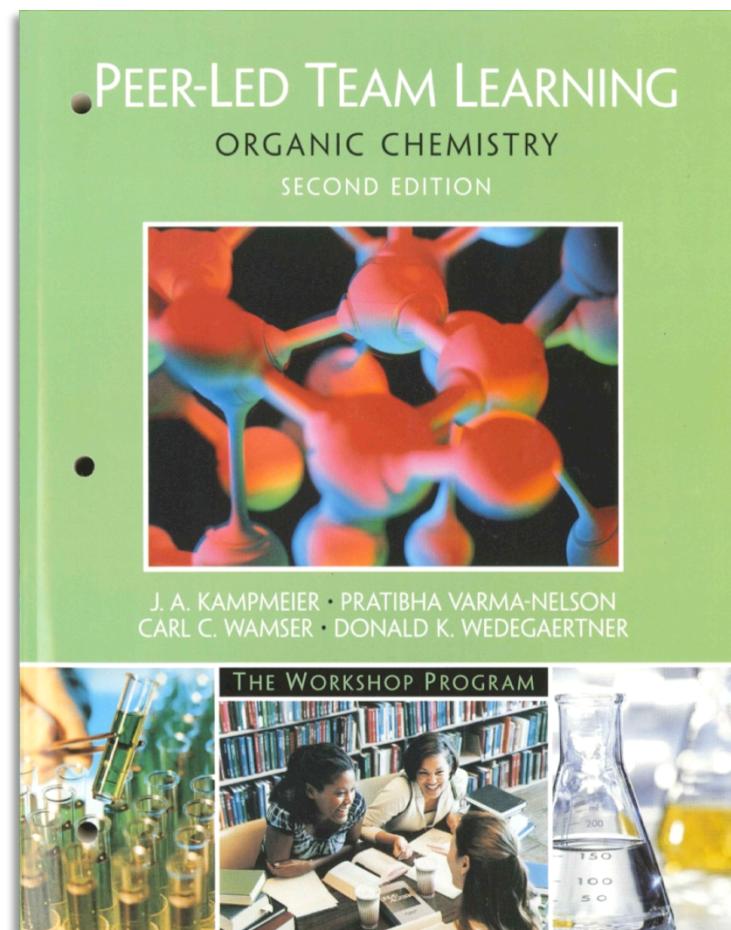
1966-69

# Textbook Writing Projects



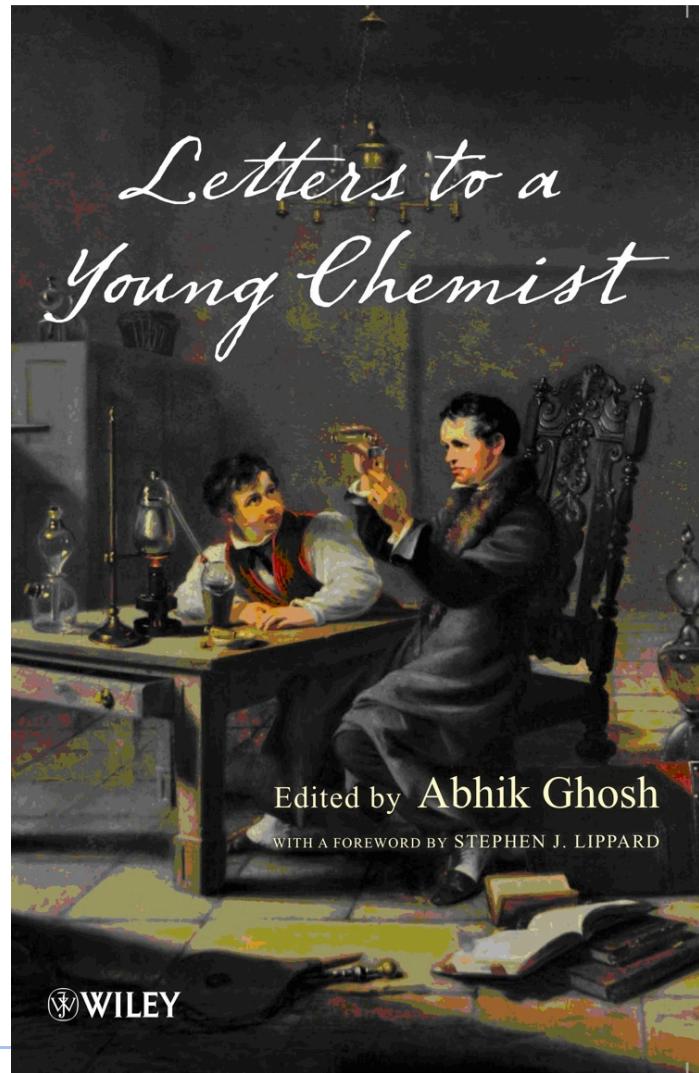
1976, 1985

# Workbook Writing Projects



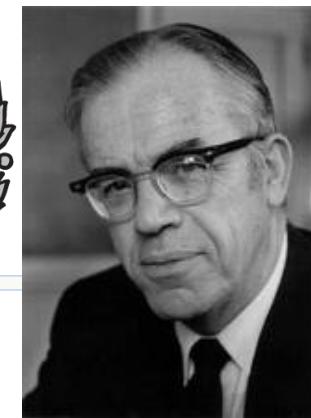
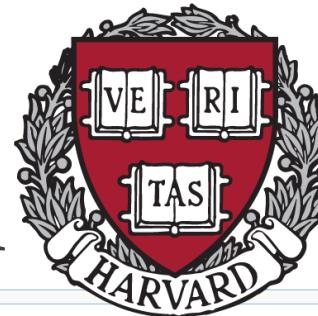
2001, 2006

# Latest Writing Project



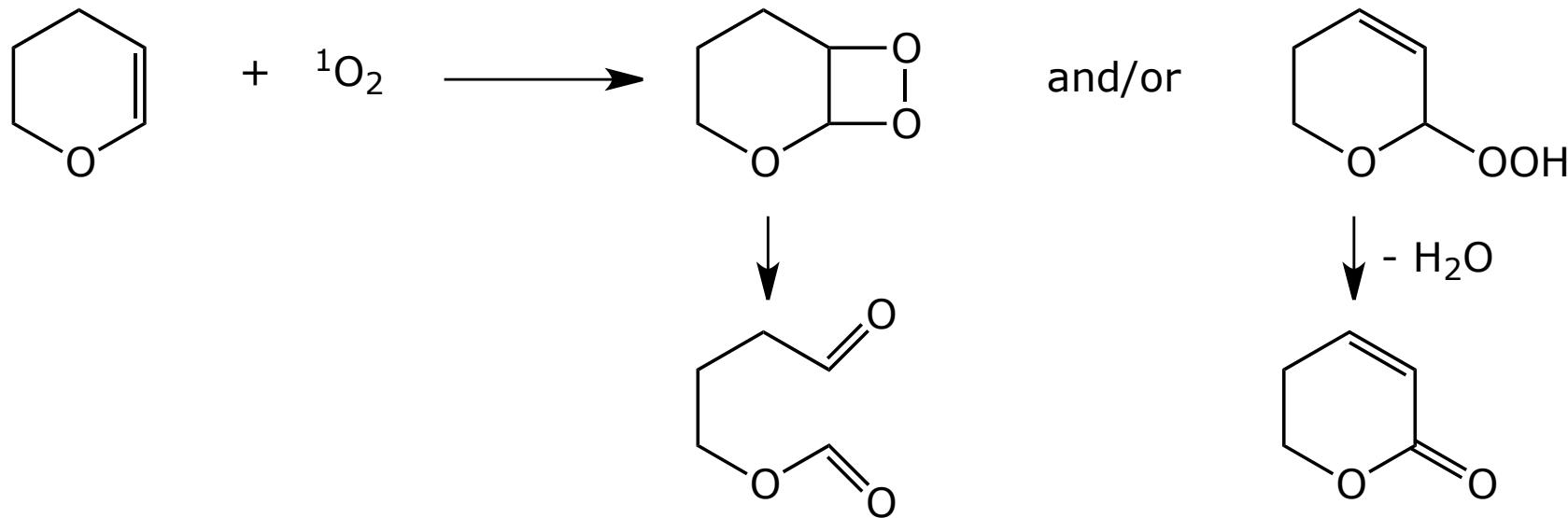
2011

# Harvard University Postdoctoral Research



## Singlet Oxygen Chemistry

Paul D. Bartlett



Reaction of Singlet Oxygen with 4-Methyl-2,3-dihydro-*g*-pyrans, A. A. Frimer,  
P. D. Bartlett, A. F. Boschung, J. G. Jewett, *J. Amer. Chem. Soc.*, 1977, 99, 7977-7986.

1969-70

# California State University, Fullerton



Andy Montana  
Department Chair



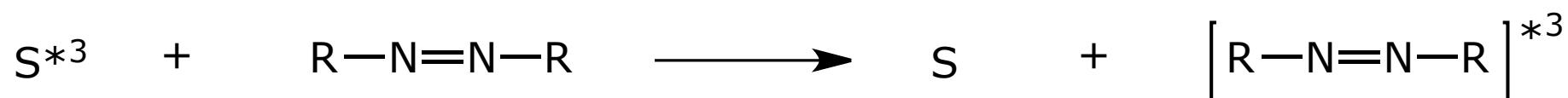
Mike King  
First Research Student



1970

# The First Papers – Energy Transfer

## Flash Photolysis

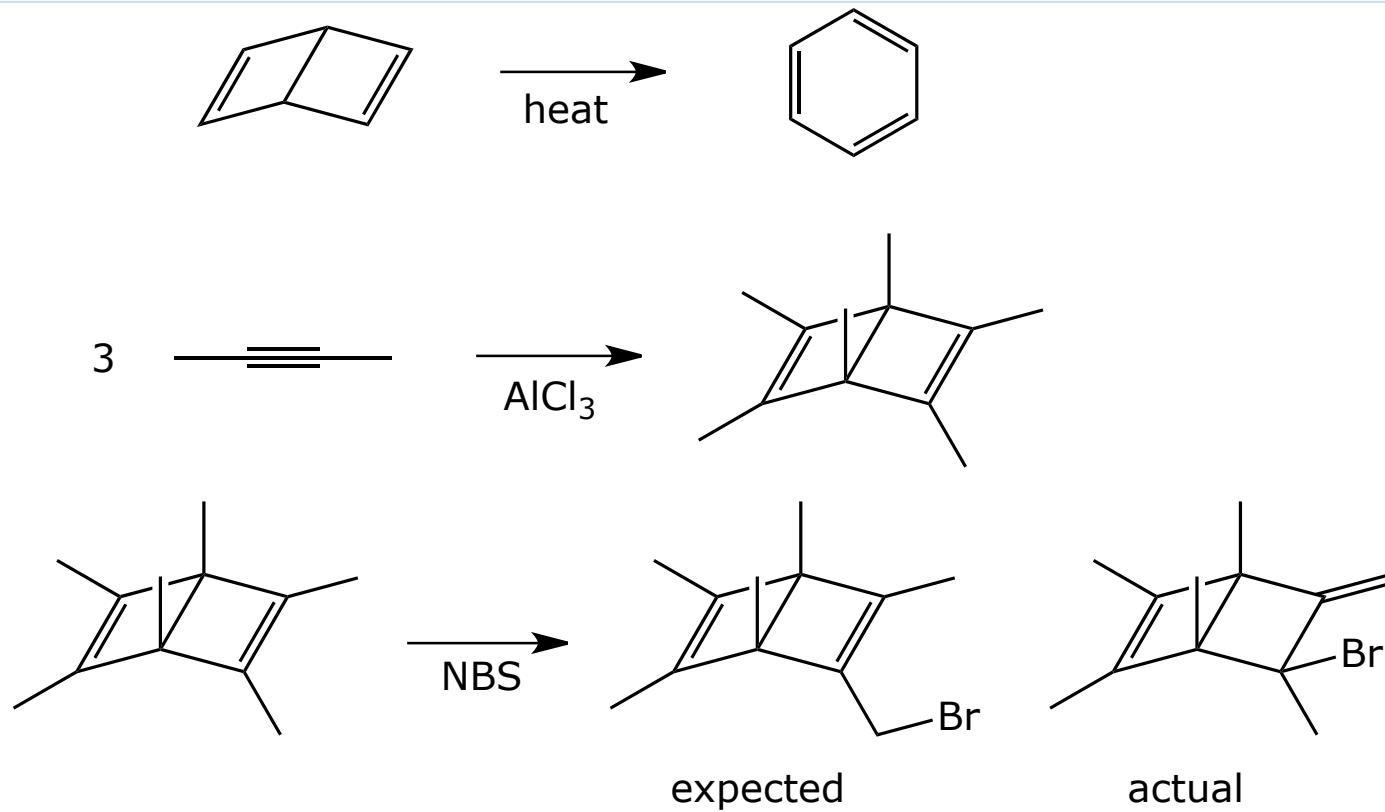


R = n-Bu, i-Bu, s-Bu, t-Bu

- **Steric Effects in Singlet and Triplet Electronic Energy Transfer,**  
C. C. Wamser and P. L. Chang, *J. Amer. Chem. Soc.* **1973**, 95, 2044-5.
- **Steric Effects in Singlet and Triplet Electronic Energy Transfer to Azo Compounds,**  
C. C. Wamser, R. T. Medary, I. E. Kochevar, N. J. Turro, and P. L. Chang,  
*J. Amer. Chem. Soc.* **1975**, 97, 4864-9.
- **Singlet Electronic Energy Transfer to Azoalkanes; Separation of Collisional and Long-Range Mechanisms by Steric and Solvent Viscosity Effects,** C. C. Wamser,  
L. Lou, J. Mendoza, and E. Olson, *J. Amer. Chem. Soc.* **1981**, 103, 7228-32.

1970s

# Dewar Benzenes



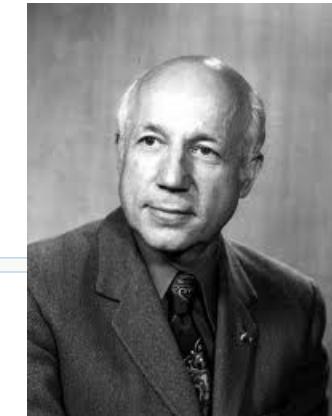
- **Hexamethyl(Dewar Benzene)**, S. A. Shama and C. C. Wamser, *Org. Synth.* **1983**, 61, 62-4.
- **Synthesis and Reactions of 5-Methylenebicyclo[2.2.0]hex-2-ene Derivatives from Hexamethyl(Dewar Benzene)**, C. C. Wamser, D. D. Ngo, M. J. Rodriguez, S. A. Shama, and T. L. Tran, *J. Amer. Chem. Soc.* **1989**, 111, 2162-2168.

1980s

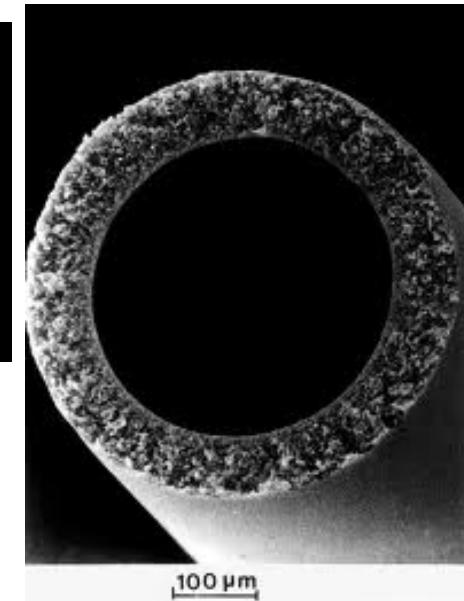
# Sabbatical, UC Berkeley



## Artificial Photosynthesis



Melvin Calvin



- Preparation and Properties of Porphyrin-Modified Hollow Fiber Membranes as Photosensitizers for Singlet Oxygen and for Artificial Photosynthesis,  
C. C. Wamser, M. Calvin, and G. Graf, *J. Membr. Sci.* **1986**, 28, 31-46.

1980

# Portland State University

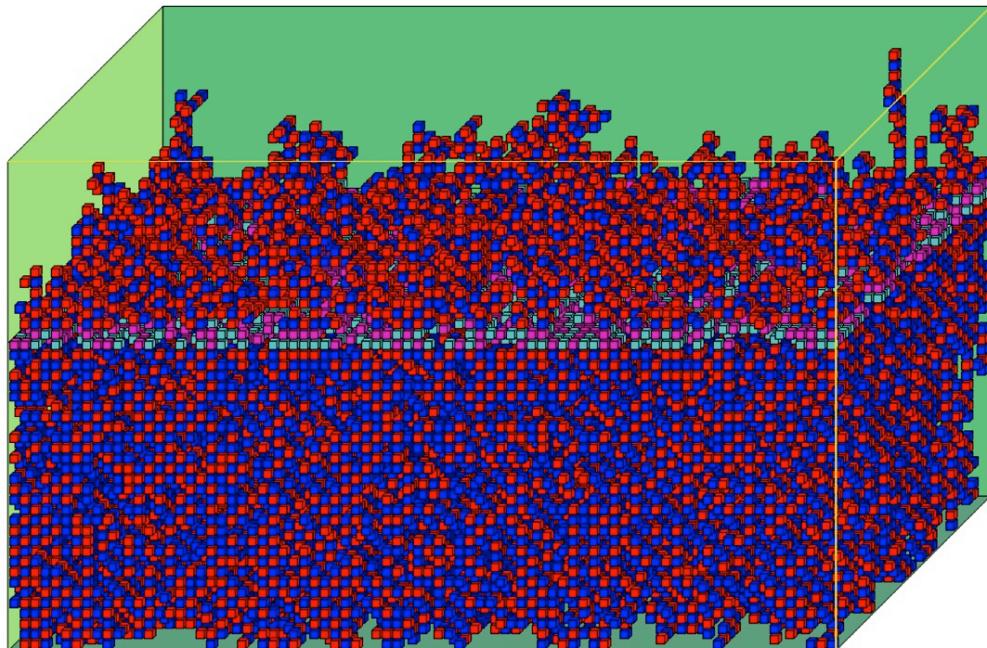


1983

# Portland State University

## Interfacial Polymerization

Bend Research, Inc.



- TAPP in  $\text{H}_2\text{O}$  (pH 3)
- TCCPP in  $\text{CH}_2\text{Cl}_2$

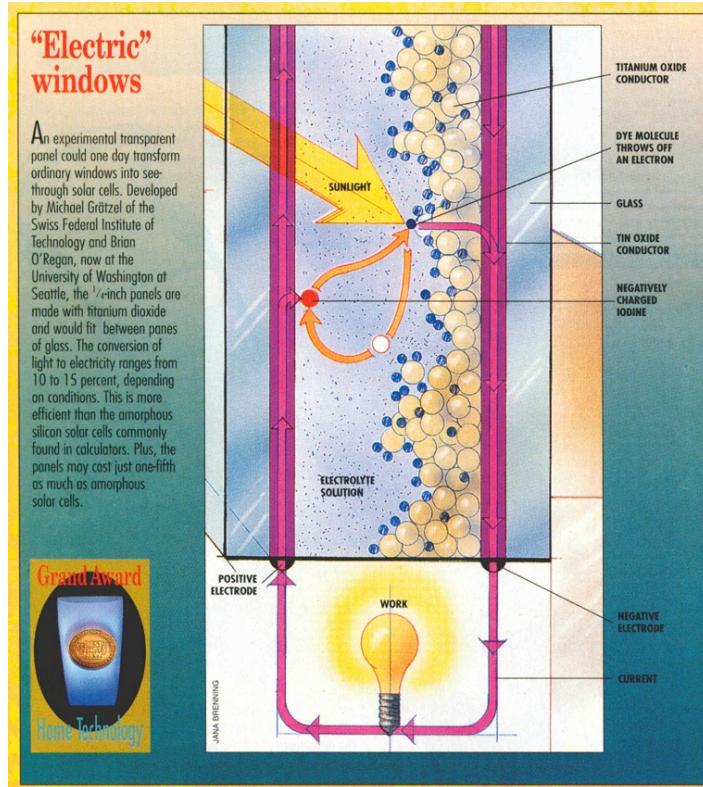
- **Synthesis and Photoactivity of Chemically Asymmetric Polymeric Porphyrin Films Made by Interfacial Polymerization**, C. C. Wamser, R. R. Bard, V. Senthilathipan, V. C. Anderson, J. A. Yates, H. K. Lonsdale, G. W. Rayfield, D. T. Friesen, D. A. Lorenz, G. C. Stangle, P. van Eikeren, D. R. Baer, R. A. Ransdell, J. H. Golbeck, W. C. Babcock, J. J. Sandberg, and S. E. Clarke, *J. Amer. Chem. Soc.* **1989**, 111, 8485-8492.

1980s

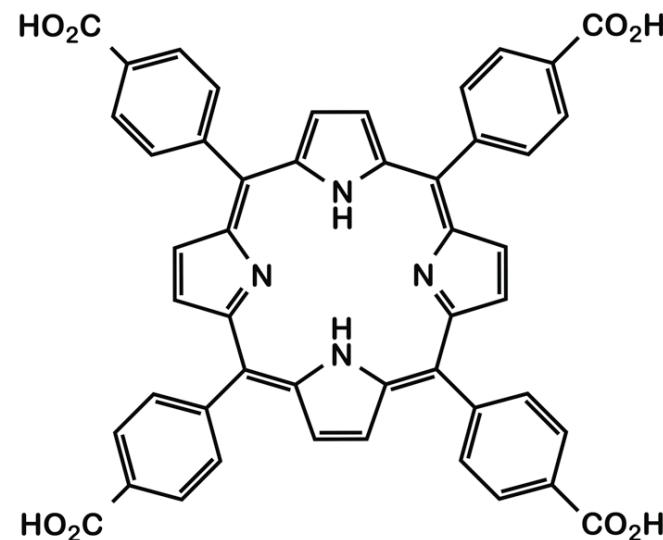
# Sabbatical, EPFL



## Dye-Sensitized Solar Cells

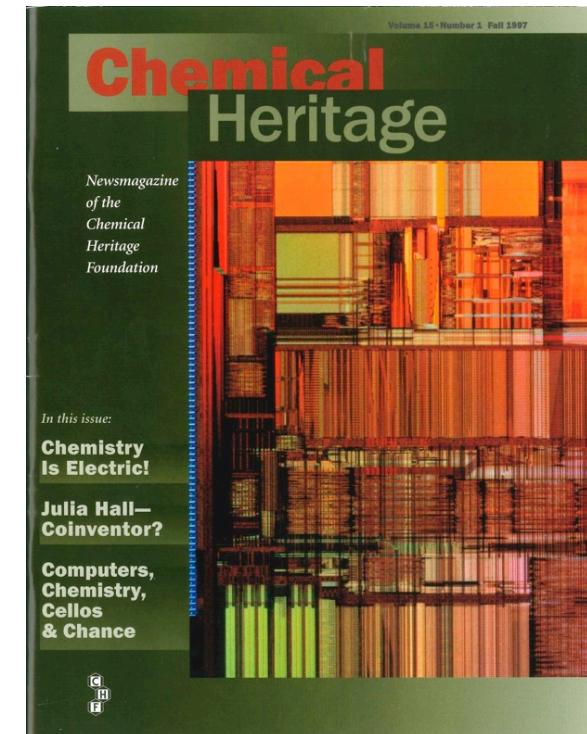
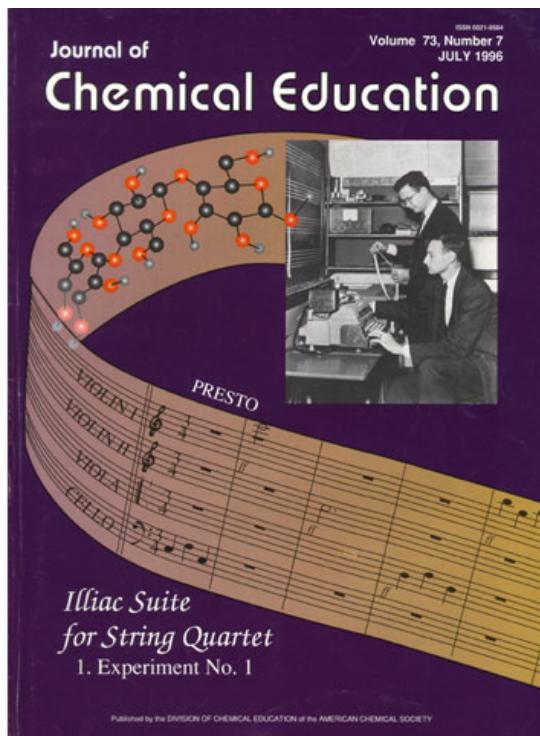


Michael Grätzel



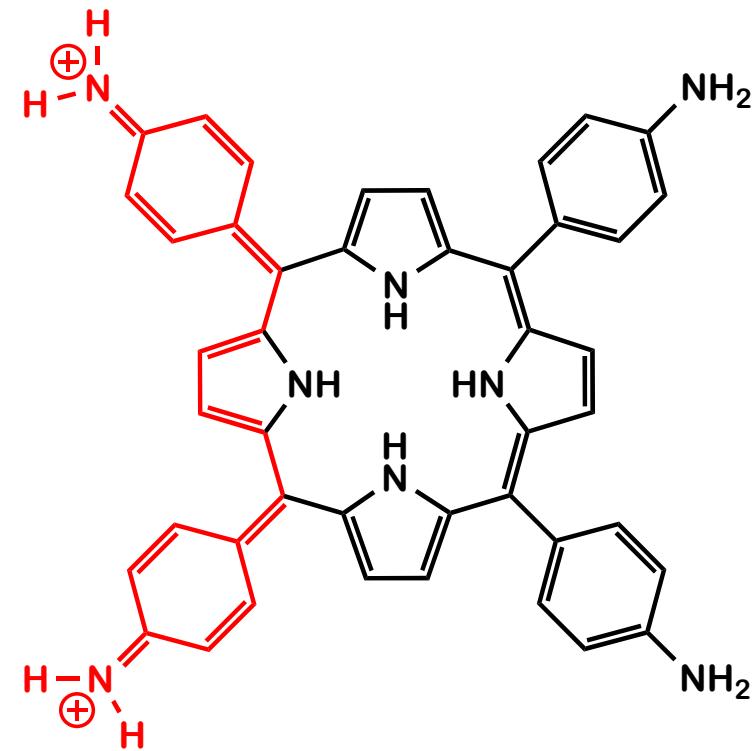
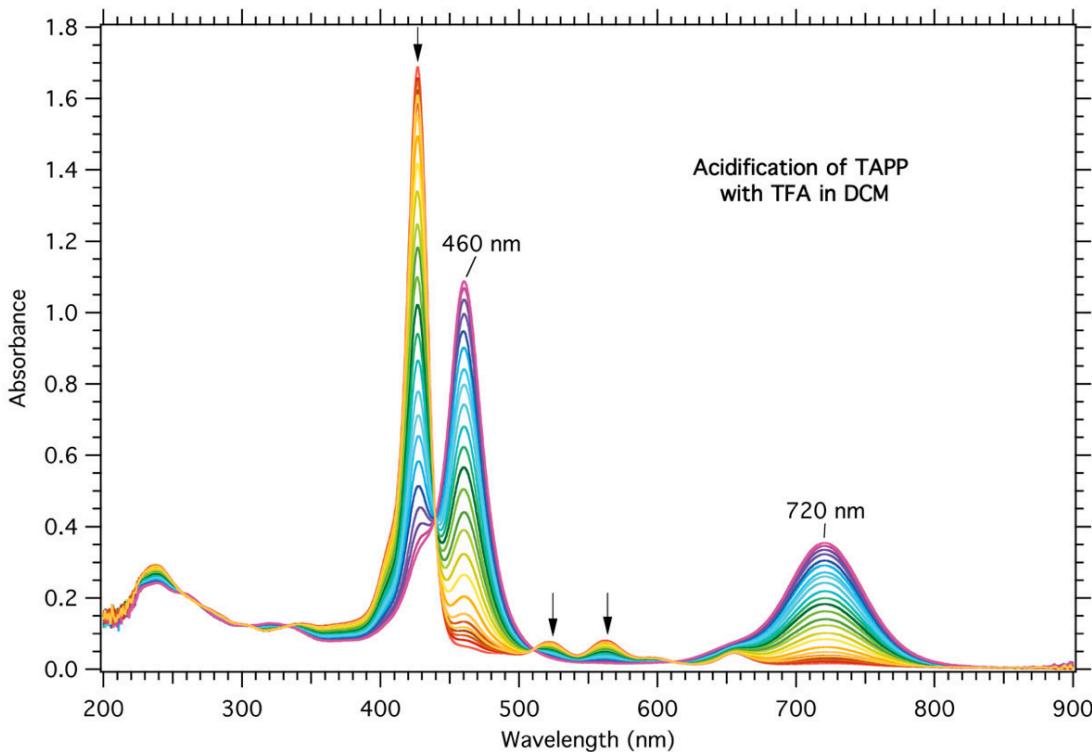
Adsorption and Photoactivity of Tetra(4-carboxyphenyl)porphyrin (TCPP) on Nano-particulate TiO<sub>2</sub>, S. Cherian and C. C. Wamser, *J. Phys. Chem. B*, 2000, 104(15), 3624-9. 1992

# Chemistry and Music: Collaborations with My Father



- **Lejaren A. Hiller, Jr.: A Memorial Tribute to a Modern Chemist-Composer**  
C. A. Wamser and C. C. Wamser, *J. Chem. Educ.* 1996, 73(7), 601-607.
- **Computers, Chemistry, Cellos, and Chance: A Celebration of Lejaren A. Hiller, Jr.**  
C. C. Wamser and C. A. Wamser, *Chem. Heritage*, 1997, 15(1), 8-9, 34-35.

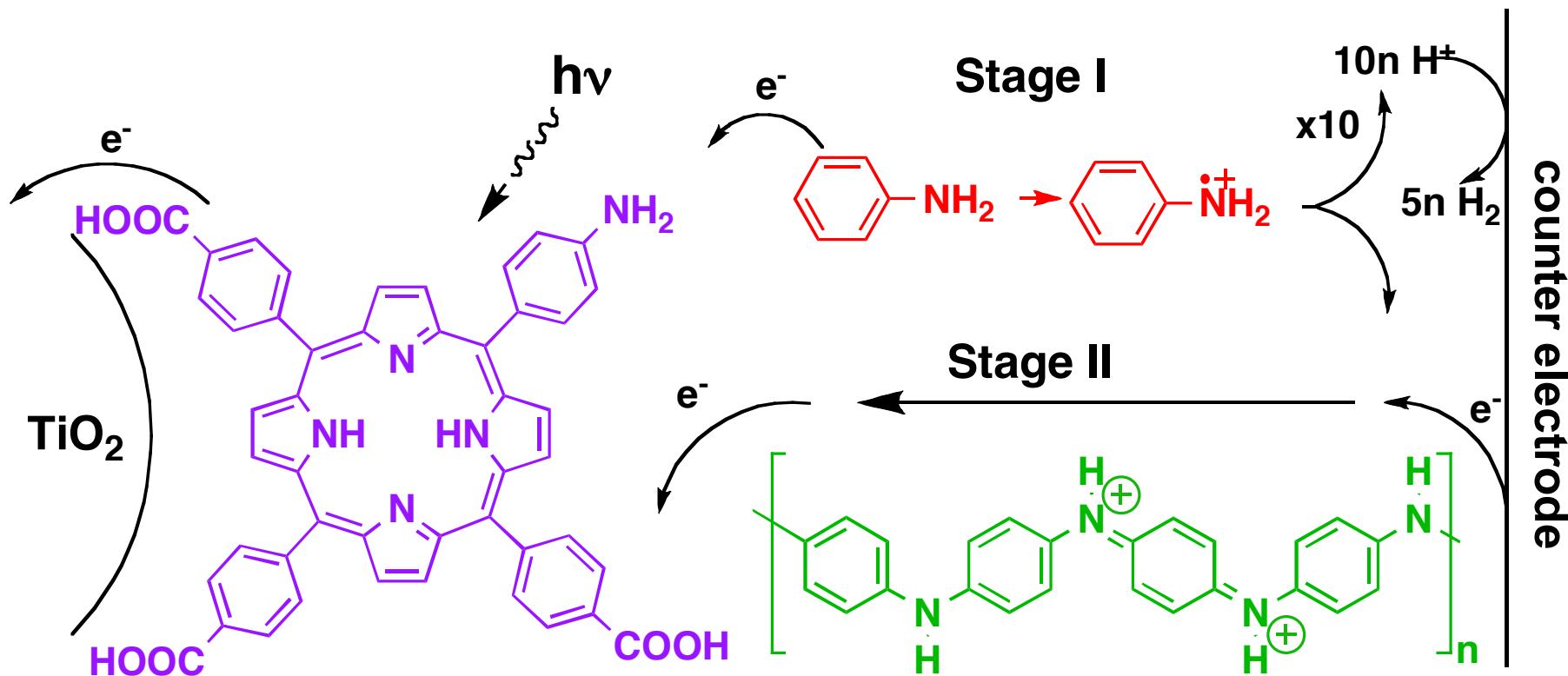
# Hyperporphyrin Spectroscopy



- **Substituent and Solvent Effects on the Hyperporphyrin Spectra of Diprotonated Tetraphenylporphyrins**, J. Weinkauf, A. Schweiger, S. Cooper, and C. C. Wamser, *J. Phys. Chem. A*, 2003, 107 (18), 3486-3496.

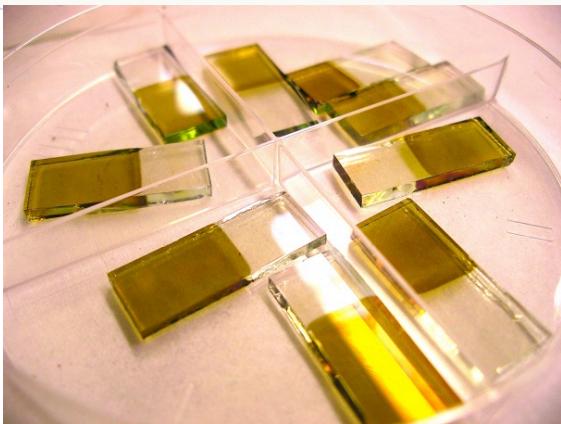
2003

# Solid-State Grätzel Cells

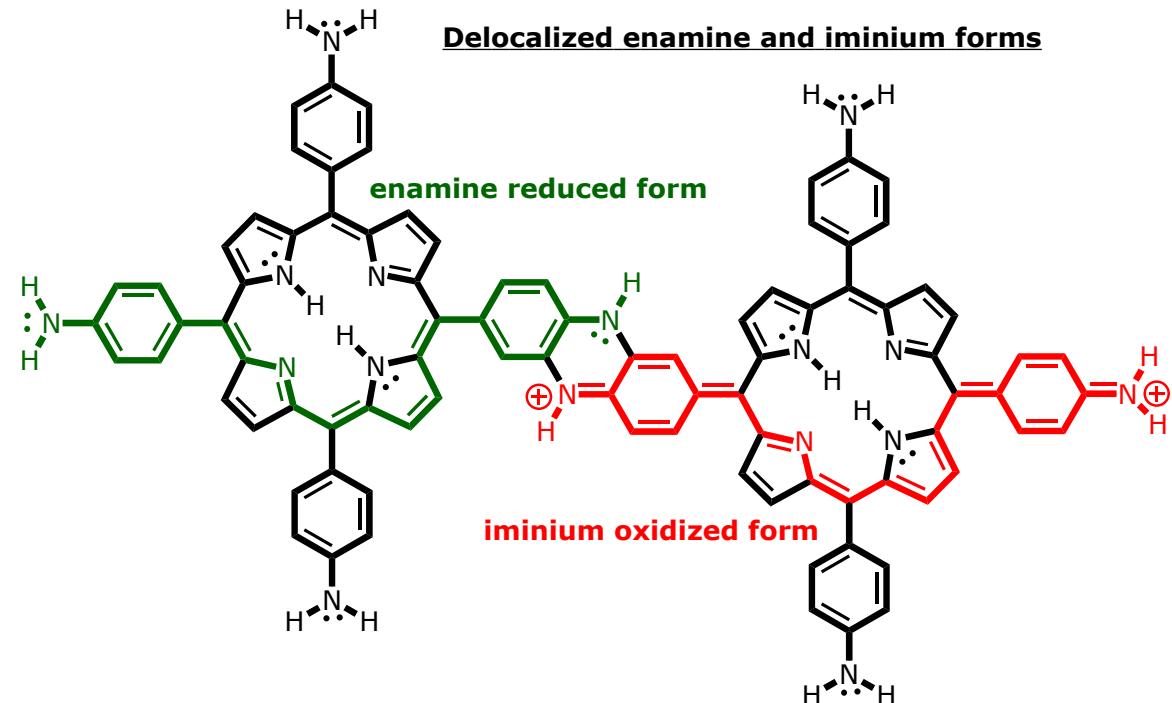
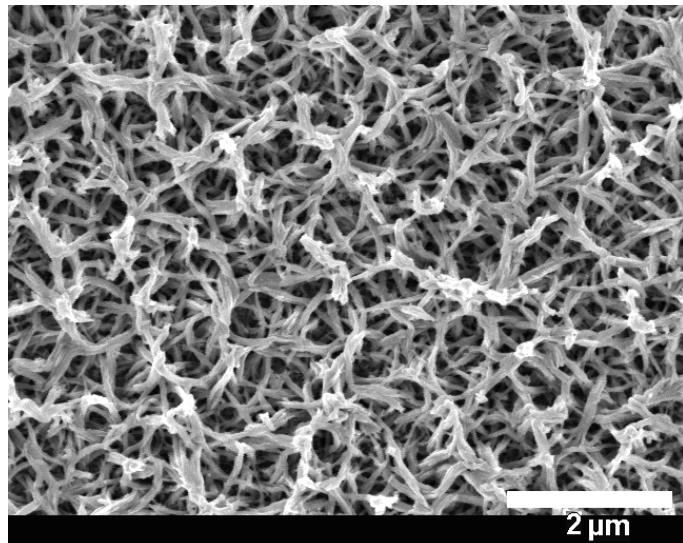


- **Photoelectropolymerization of Aniline in a Dye-Sensitized Solar Cell,**  
H.-S. Kim and C. C. Wamser, *Photochem. Photobiol. Sci.*, **2006**, 5 (10), 955-960.

# Inverse Grätzel Cells



*Poly-TAPP*



**Synthesis and Characterization of Electropolymerized Nanostructured Aminophenylporphyrin Films, M. G. Walter and C. C. Wamser, *J. Phys. Chem. C*, 2010, 114, 7563-7574.**

2010

# The Journal of Organic Chemistry

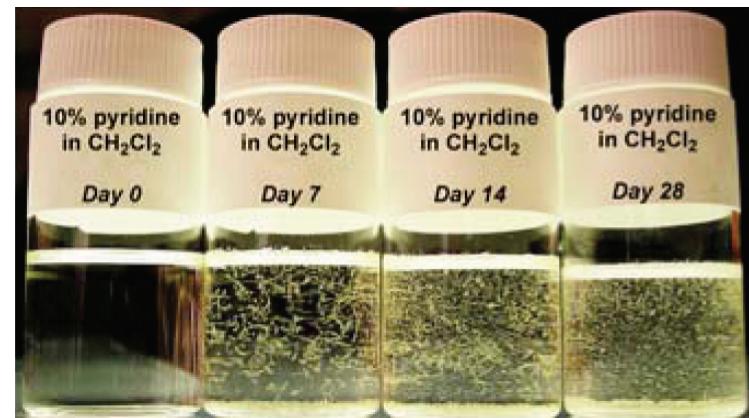
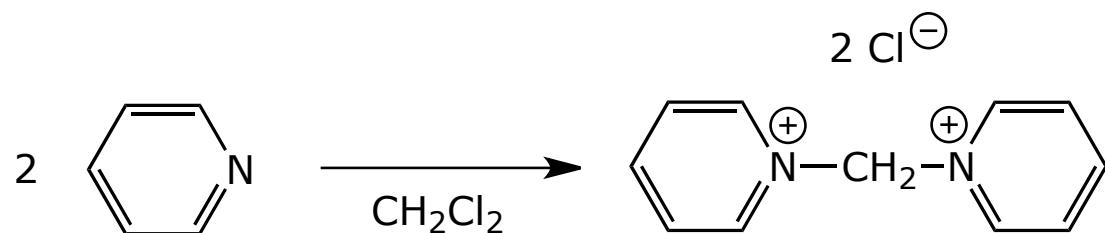
Listed below are the most accessed *JOC* articles during the second quarter of 2010:

## Reaction of Dichloromethane with Pyridine Derivatives under Ambient Conditions

*Alexander B. Rudine, Michael G. Walter, and Carl C. Wamser*

*J. Org. Chem.*, 2010, 75 (12), pp 4292–4295

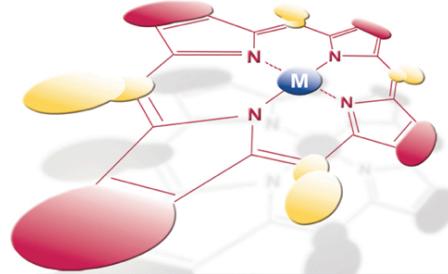
DOI: 10.1021/jo100276m



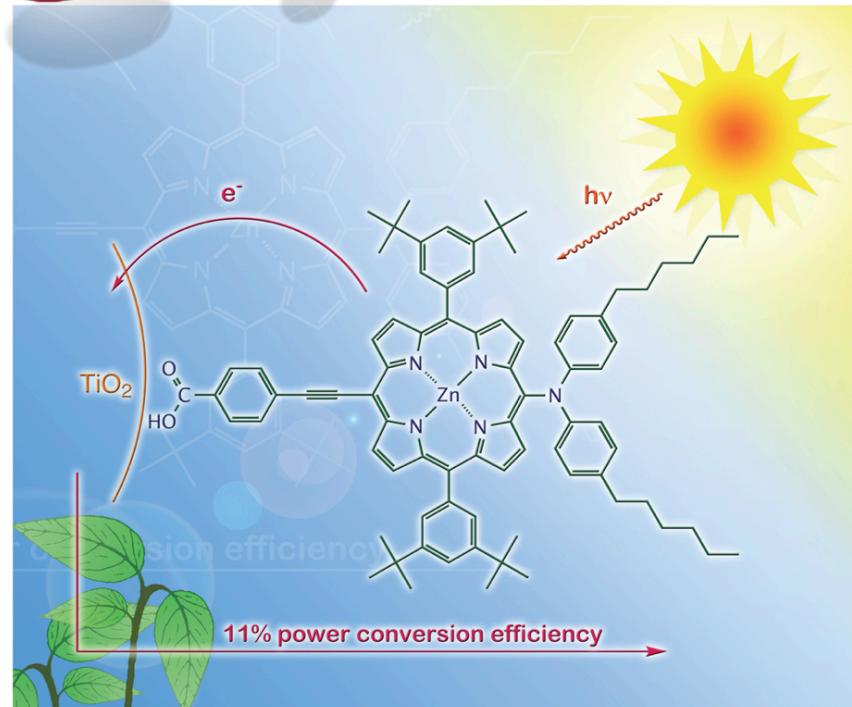
2010

The official journal of SPP

# Journal of Porphyrins and Phthalocyanines



An international journal  
devoted to research in the  
chemistry, physics, biology  
and technology of porphyrins,  
phthalocyanines and related  
macrocycles



[www.worldscinet.com/jpp/](http://www.worldscinet.com/jpp/)

Society of  
Porphyrins &  
Phthalocyanines

# Review Article

## Porphyrins and Phthalocyanines in Solar Photovoltaic Cells

Alexander B. Rudine,  
Michael G. Walter, and  
Carl C. Wamser

*J. Porphyrins Phthalocyanines,*  
**2010; 14: 759–792.**

2010

# Sabbatical Caltech

Polymer Solar Cells



Nate Lewis



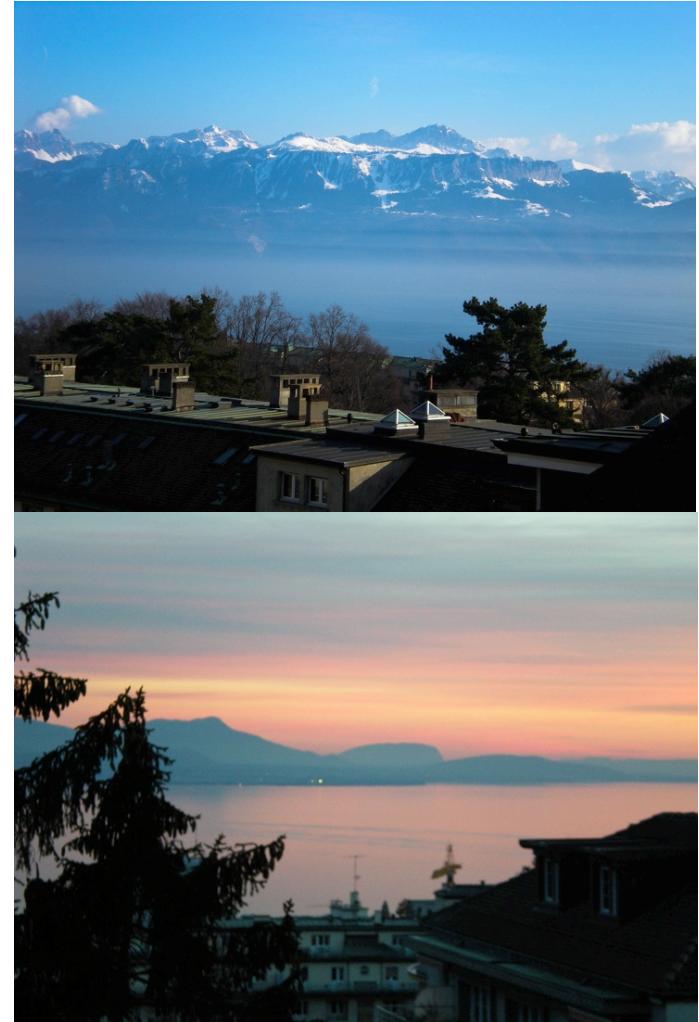
2003



Moscow, 2008

# Sabbatical EPFL+

Lausanne, Switzerland



2011

# Sabbatical EPFL+

University of Nantes, France



Fabrice Odobel



2011

# Sabbatical EPFL+

Valencia, Spain



3rd HYBRID AND ORGANIC PHOTOVOLTAICS CONFERENCE  
**HOPV2011**  
15 - 18 MAY 2011 - VALENCIA - SPAIN



2011

# Sabbatical EPFL+

Italy - Milan, Pavia, Venice



2011

# Sabbatical EPFL+

University of Tromso, Norway



Abhik Ghosh



2011

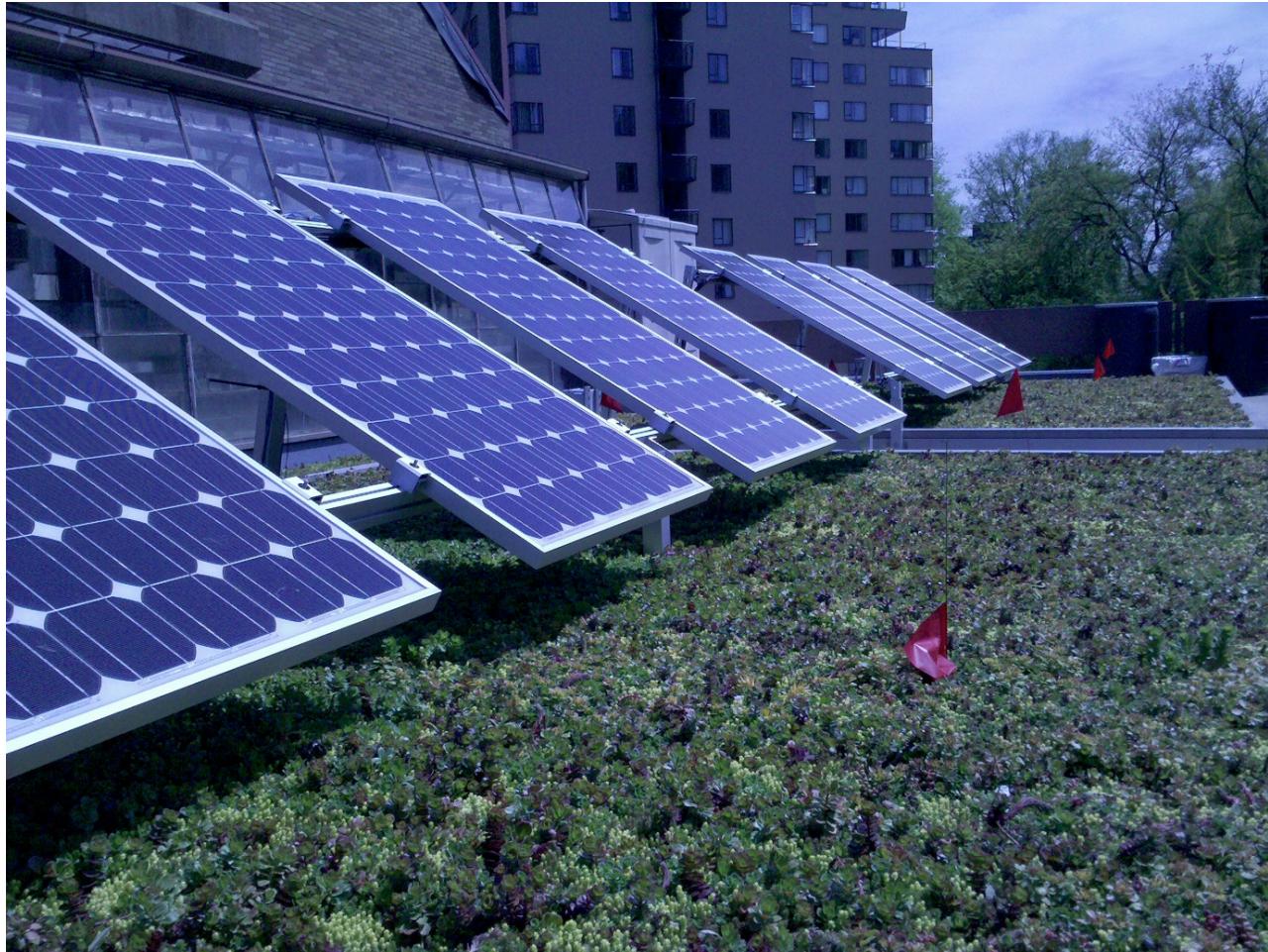
# Sabbatical EPFL+

University of Edinburgh, Scotland



2011

# PSU Photovoltaic Test Facility



Co-PIs:

David Sailor  
Todd Rosenstiel  
(Erik Johansson)

Sponsors:

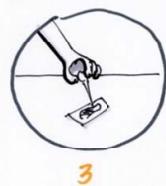
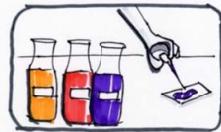
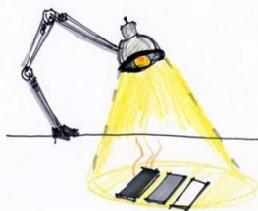
NSF  
PGE  
OR BEST  
PSU ISS  
Portland BES  
PSU OGSR  
EC Company  
UO SRML

# NSF - CRPA

OMSI

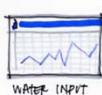
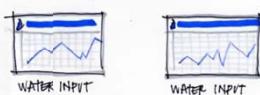
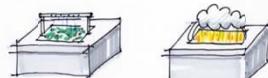
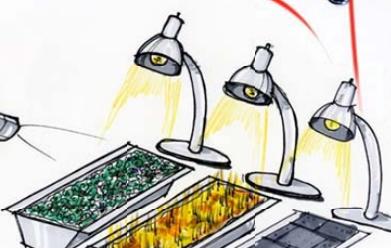
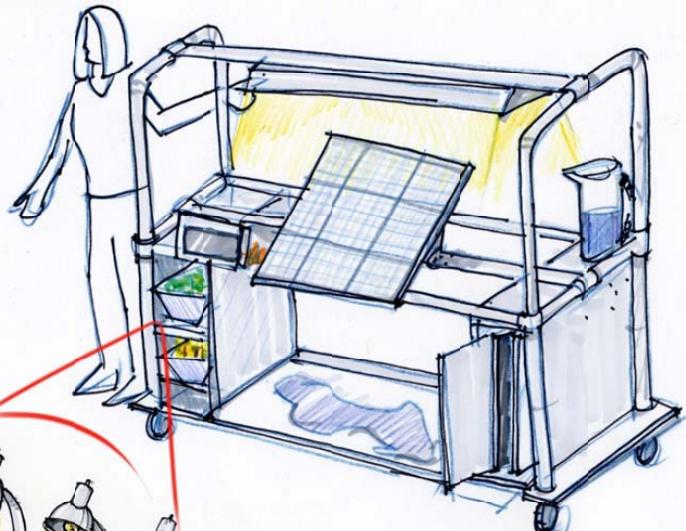
## Connecting Researchers to Public Audiences

WHAT DYE IS BEST?

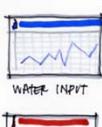
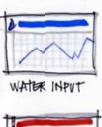
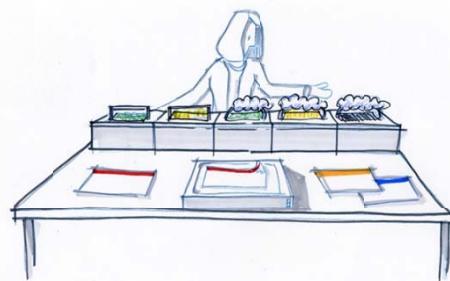


3 4

WATT'S ON THE ROOF



DATA ANALYSIS OF GREEN ROOF VARIABLES



# Humanity's Top Ten Problems for the next 50 years

1. ENERGY
2. WATER
3. FOOD
4. ENVIRONMENT
5. POVERTY
6. TERRORISM & WAR
7. DISEASE
8. EDUCATION
9. DEMOCRACY
10. POPULATION



Partners in Science    R. E. Smalley  
January 18, 2003    Rice University

# O Grignard

The carbonyl is polarized  
The carbon end is plus.  
The nucleophile will thus attack  
The carbon nucleus.  
The Grignard makes an alcohol,  
Of types there are but three.  
It makes a bond  
To correspond  
From C to shining C.

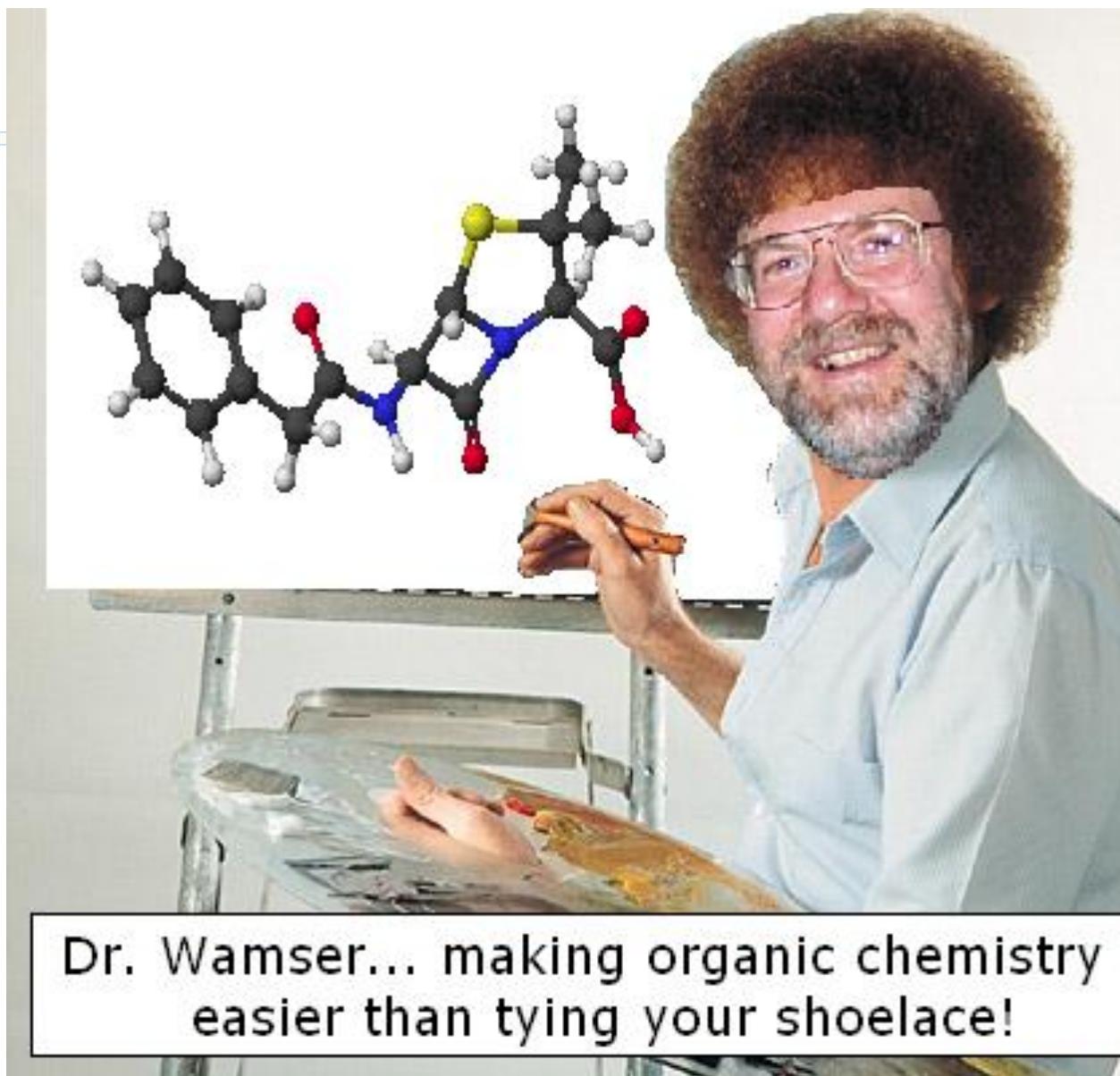
Lyrics by Frank Westheimer (Harvard Univ)





# CARL of the (CARBON) RINGS

THE MESO TOWERS



Dr. Wamser... making organic chemistry  
easier than tying your shoelace!

# Organic Chemistry – Spring 2012



