

Where do Minerals concentrate *within* the crust? (Ch.2)

1. Soil (through weathering)
2. Groundwater
3. Magmatic / Metamorphic processes



Oregon Industrial Mineral Producers

(exclusive of aggregate and gemstones)

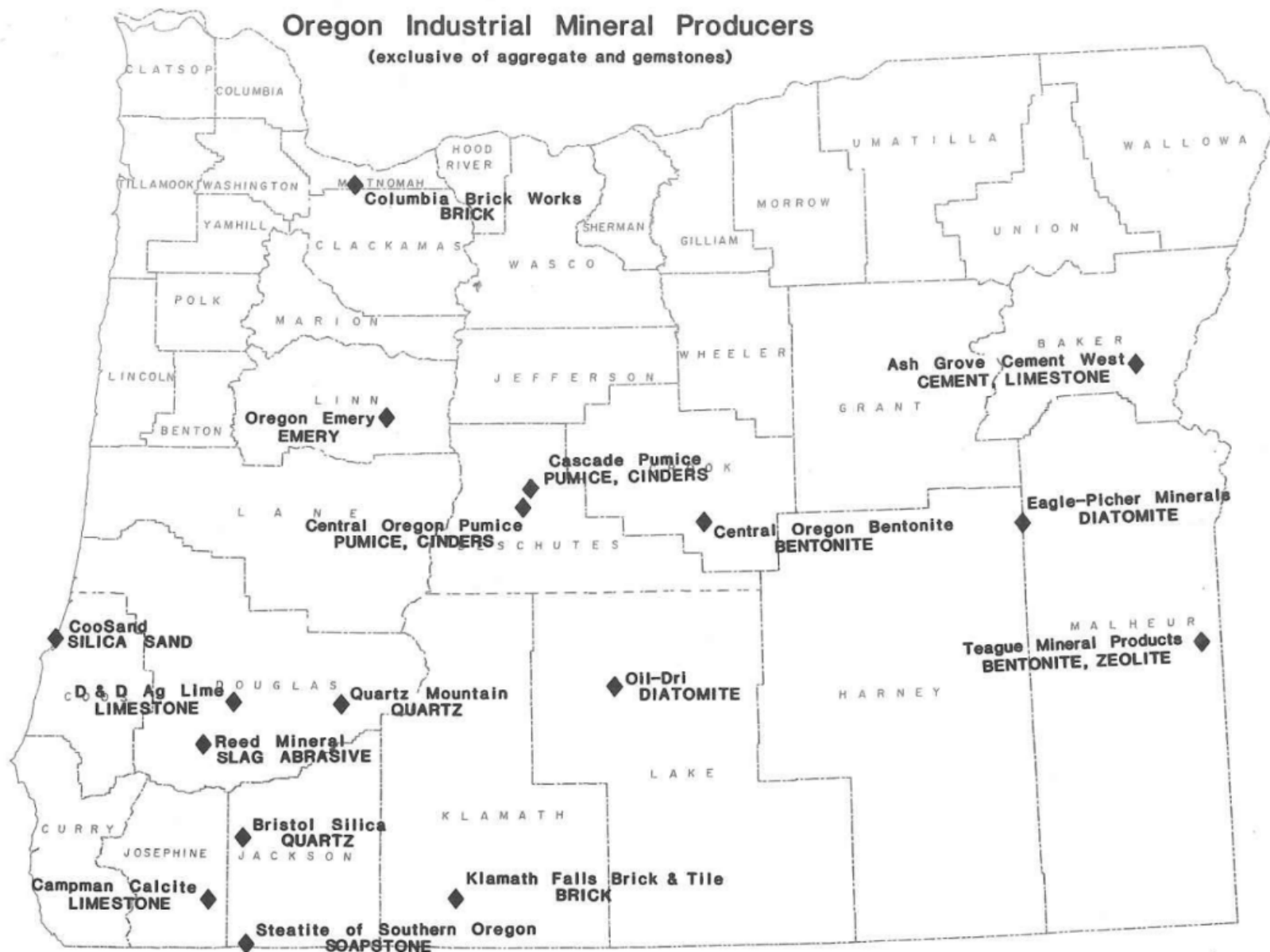
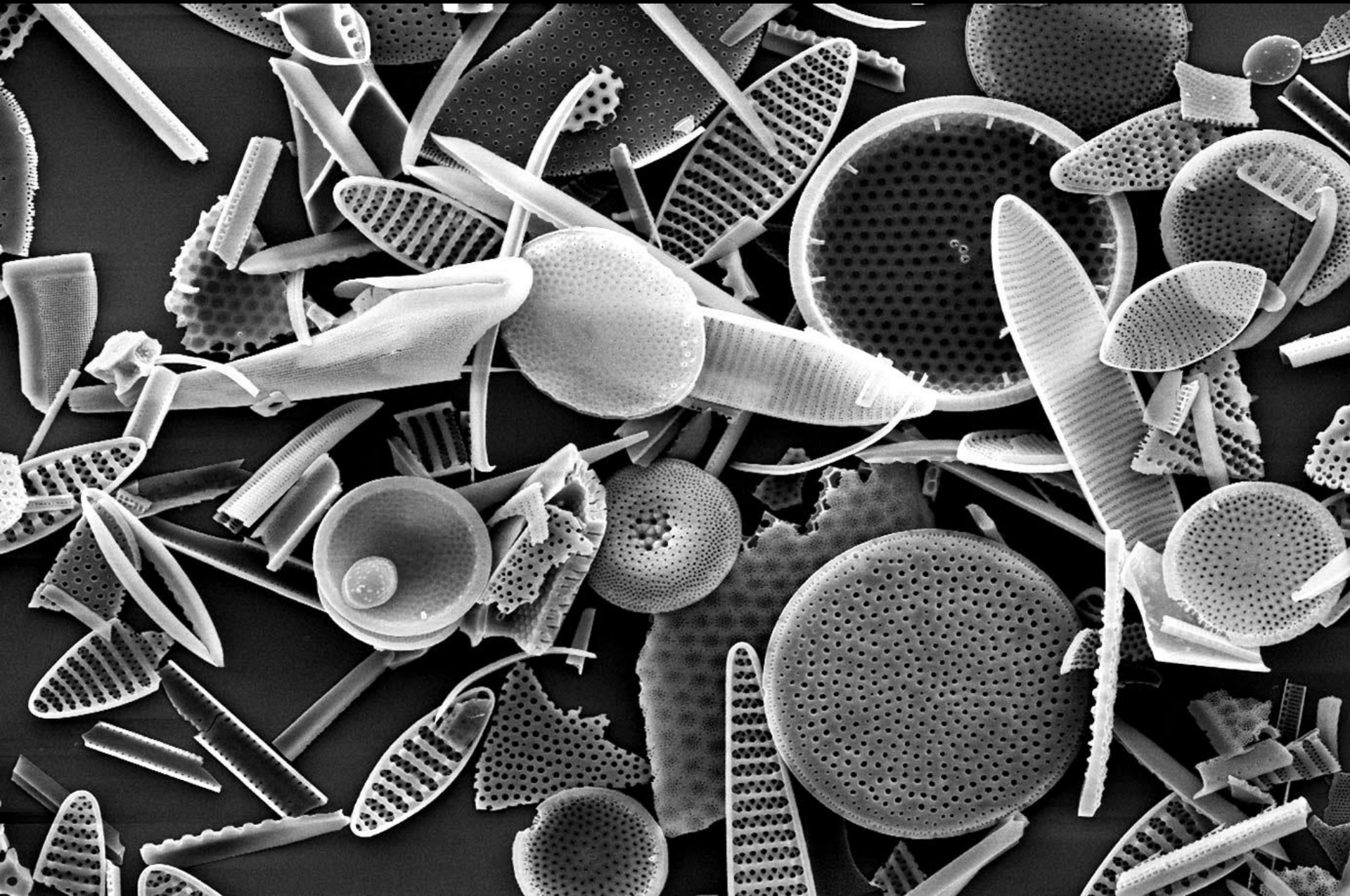


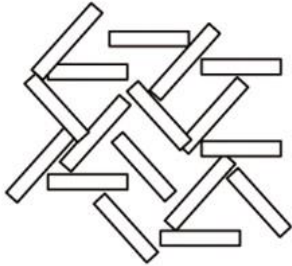
Figure 1. Current producers of industrial minerals in Oregon.

Diatoms (microfossils with silica shells)

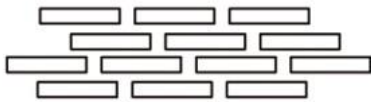


SHALE

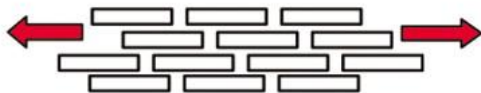
1. Deposition



2. "Collapse"

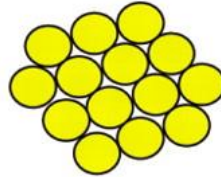


3. Interlayer water release

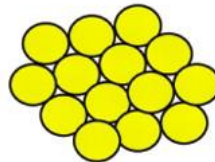


SAND

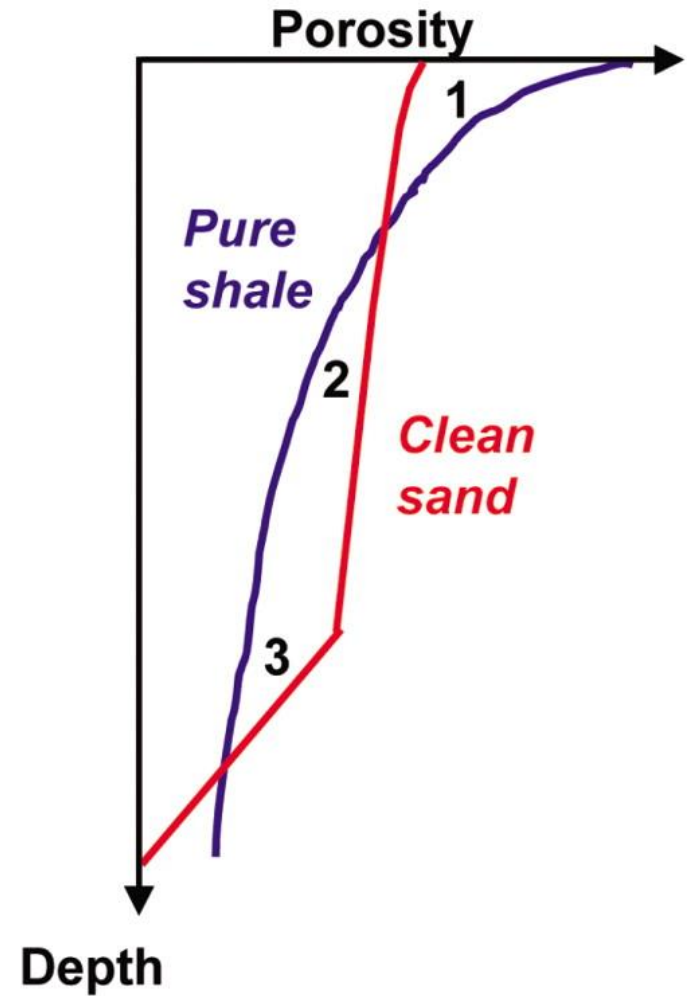
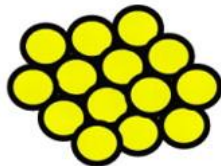
1. Deposition

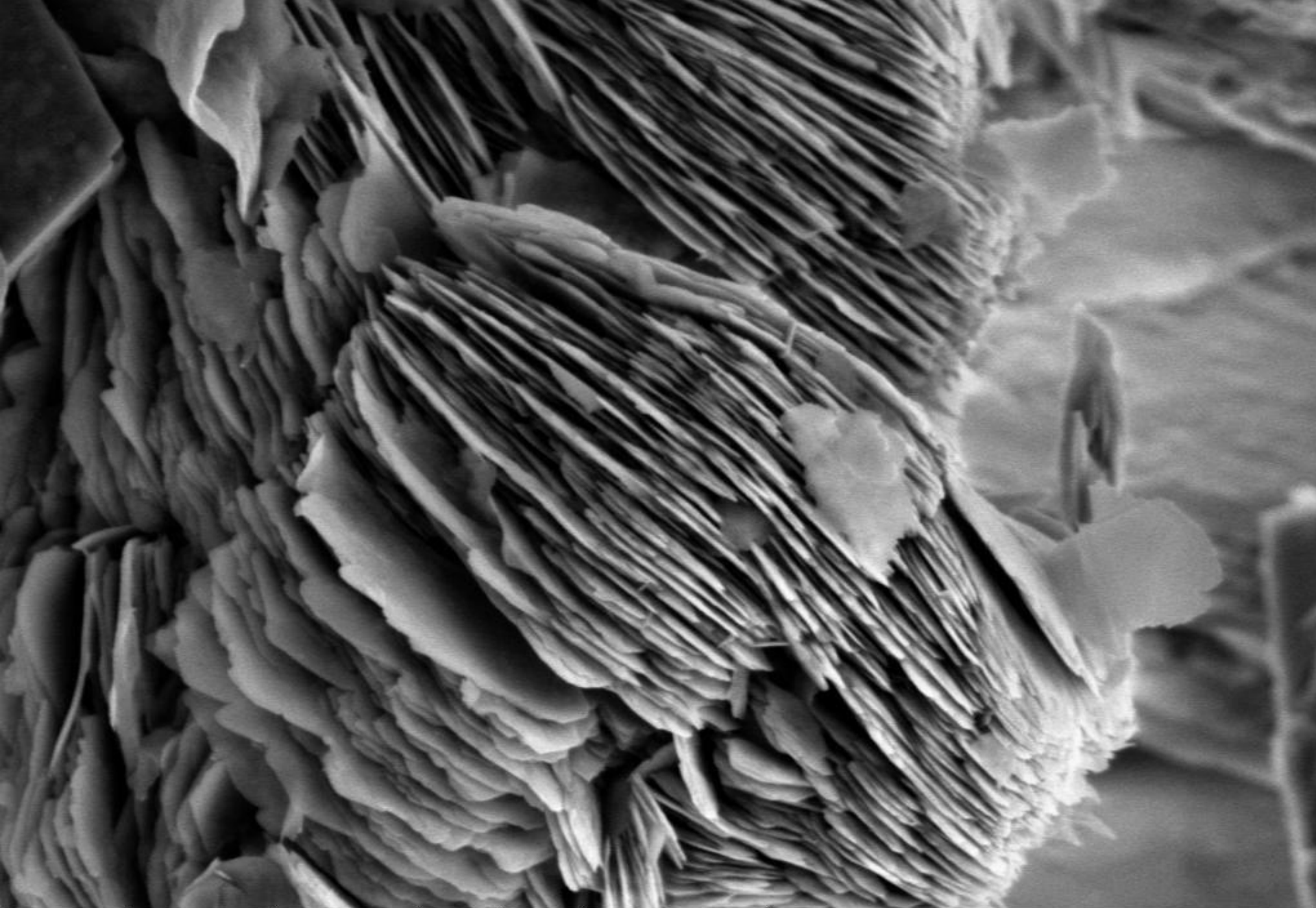


2. Compaction



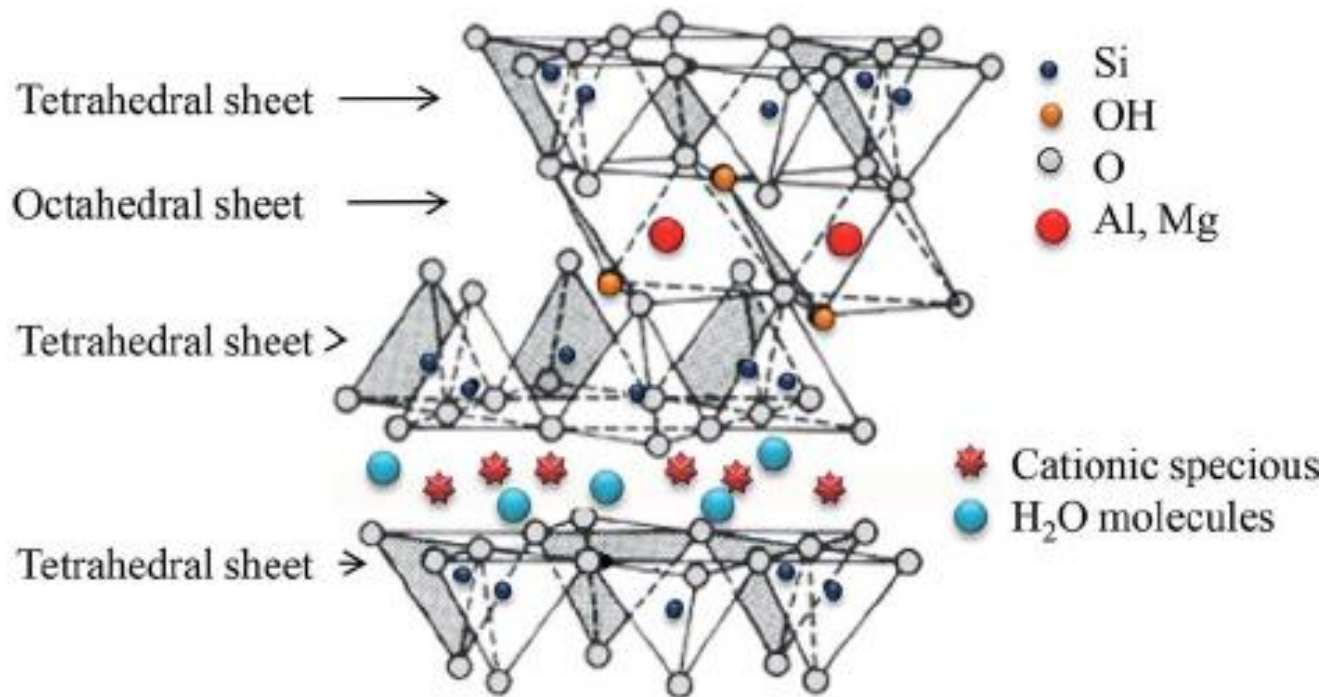
3. Cementation





HV	mag □	WD	spot	det	3/29/2011	5 μm
5.00 kV	24 000 x	11.5 mm	3.0	ETD	1:03:44 PM	The University of Jordan - SEM Unit

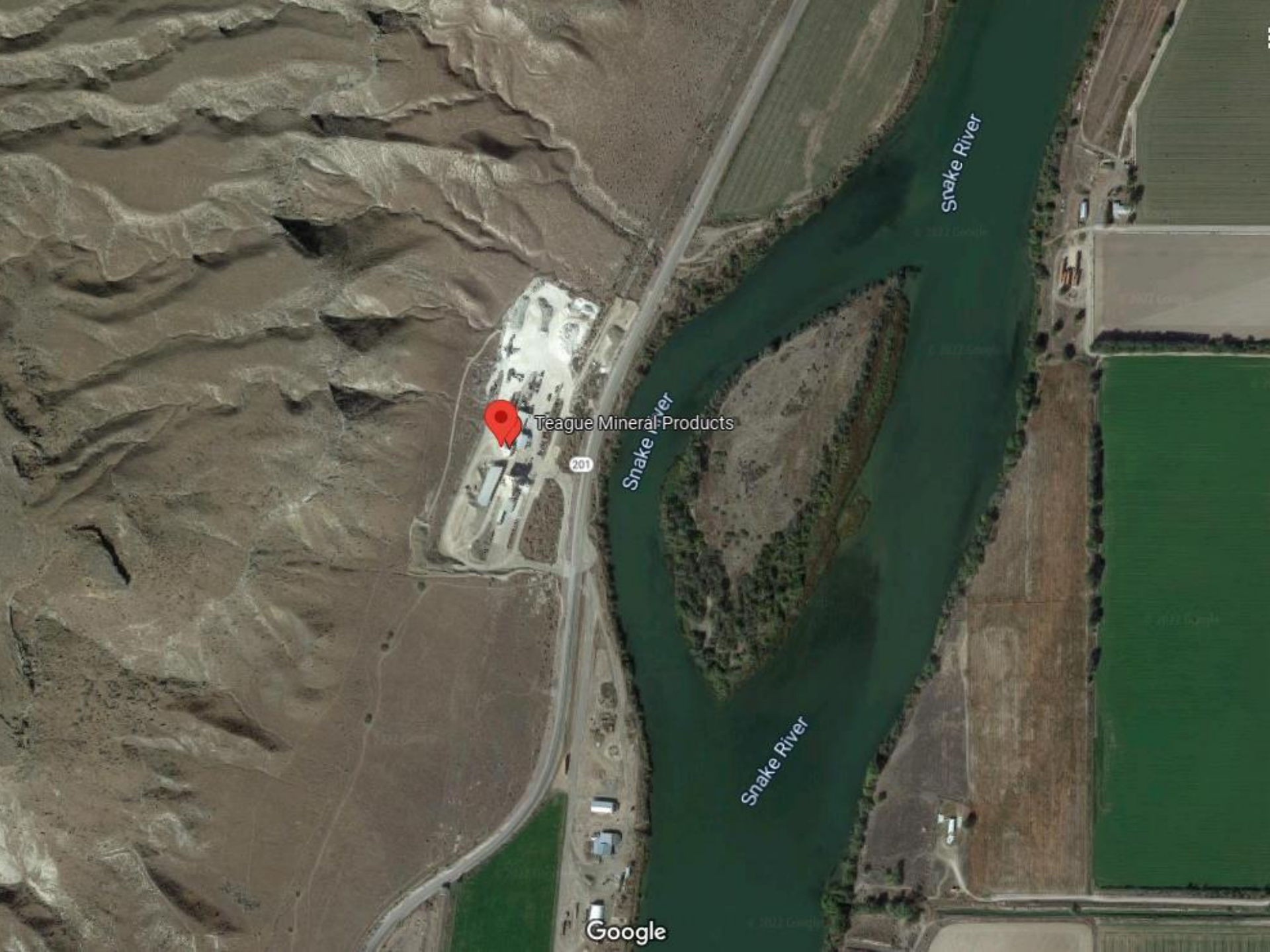
Clay mineral structure (generalized)





Kaolin(ite) →





Snake River

Snake River

Snake River

Teague Mineral Products

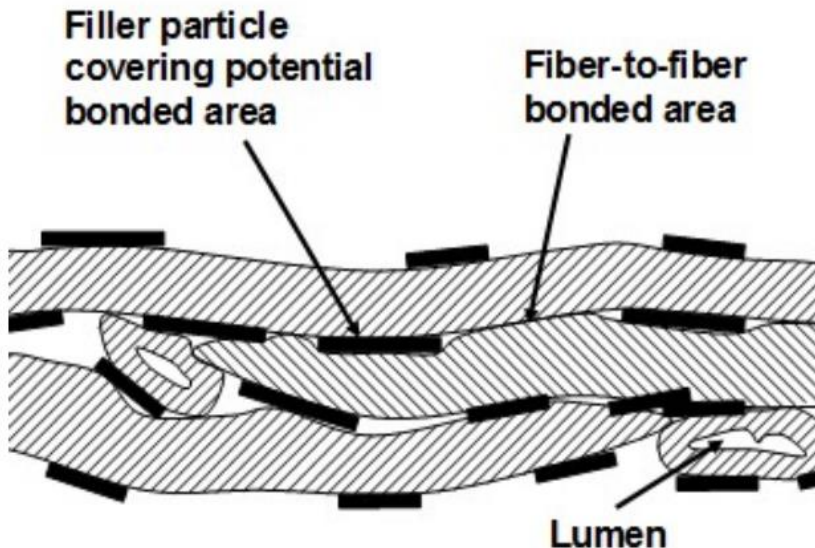
201

Google

Clay, what's it good for?

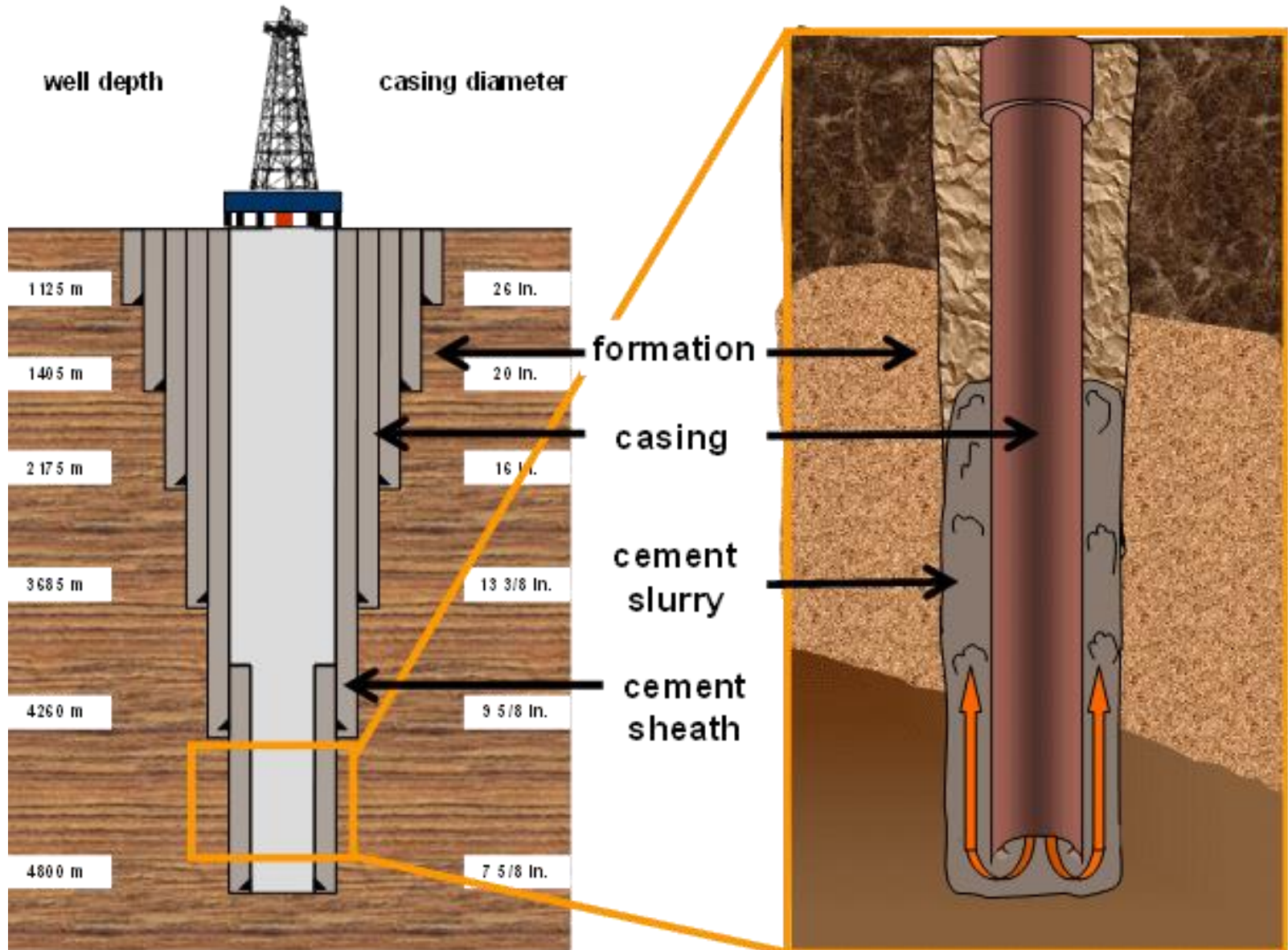
Absorption
(spills, drilling, etc.)

Paper!!



Benefits of Filler Use	Ways to Maximize the Benefits	Citations
Cost reduction	Increased filler content, using low-cost fillers	Davidson 1965
Brightness	Increased filler content, using bright fillers	Bauch 1992
Opacity	Increased filler content, especially using fillers with optimum size of the primary particles. Best results with high refractive index filler	Bown 1998
Reduced energy of drying the paper	Increased filler content	Kapoor <i>et al.</i> 1996
Friction – to increase the paper-to-paper coefficient	Composite structure of filler, e.g. calcined clay, scalenohedral PCC	Rooks 1993
Friction – to decrease the paper-to-paper coefficient	Platy filler with easy delamination, e.g. talc	Biza 1999
Control of pore size, e.g. for filter papers	Selection of size and particle shape of filler	Schwalbe 1962
Controlled burn rate, e.g. for cigarette paper	Use of aluminum trihydrate (ATH)	Schwalbe 1962

Clay is used primarily for drilling (mud)



Multifunctional Clay in Pharmaceuticals

Abstract

Clay has its widespread applications in pharmaceuticals from ancient world to modern era. It is one of the excellent excipients present in the commercially available pharmaceuticals. Its use in many of dosage forms viz. in suspension, emulsion, ointments, gels, tablet and as drug delivery carrier as suspending agent, emulsifying agent, stiffening agent, binder, diluent, opacifier, and as release retardant have been explored in many studies. Variety of minerals is used as both excipient and as an active ingredient; among that kaolinite, talc, and gypsum are important. Their inertness, low toxicity, versatile physiochemical properties and cost effectiveness has increased its usage in pharma industries. Many minerals have its own pharmacological action as antacid, anti-bacterial, anti-emetic, anti-diarrheal agent and as skin protectant etc. Their unique structure which helps them to absorb material onto their layered sheets has opened a wide variety of applications in drug delivery. The understanding of surface chemistry and particle size distribution of clay minerals has led the pharmaceutical field in many directions and future perspectives.

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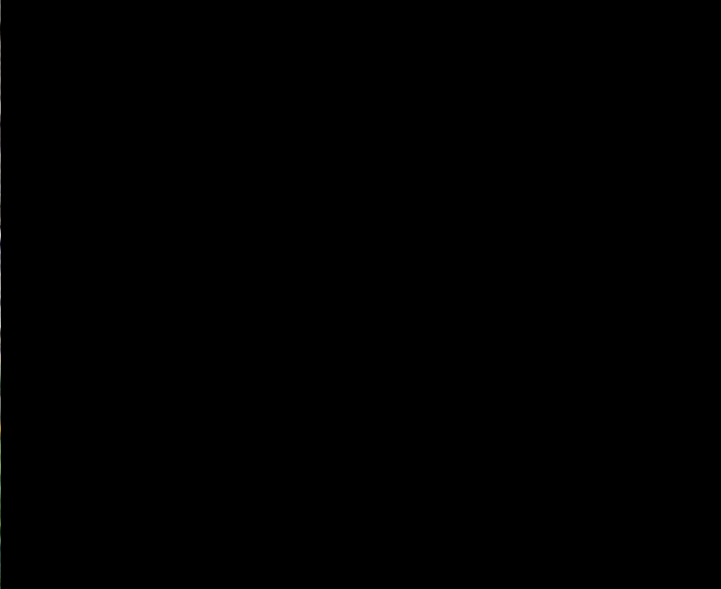
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Gypsum is an “evaporite”
Salts form when lakes dry up



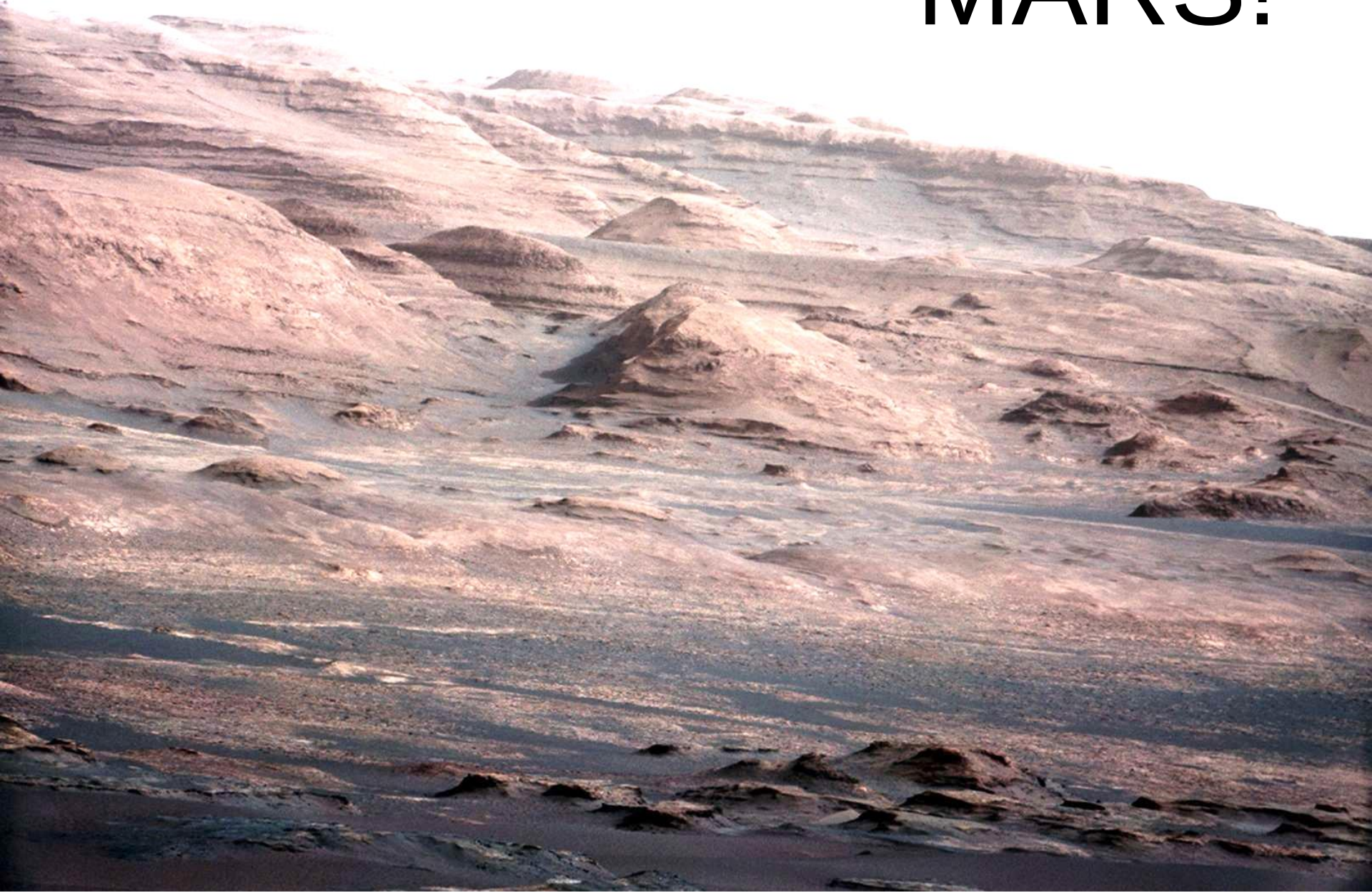
Empire Mining Company (Gypsum)

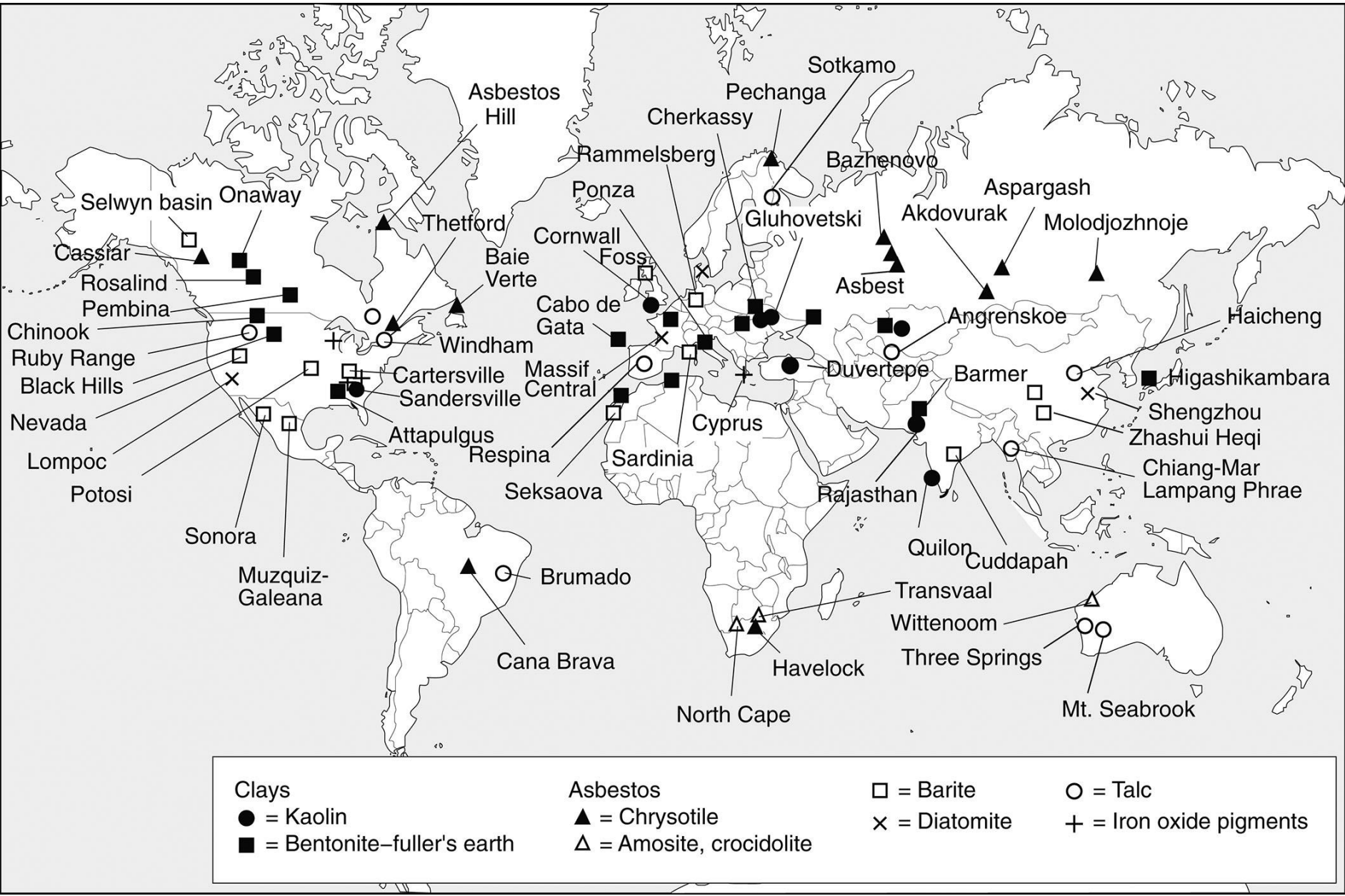


“Tufa Towers” at Mono Lake
Calcium Carbonate (CaCO_3) = Calcite!

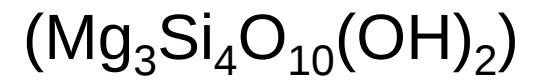


MARS!



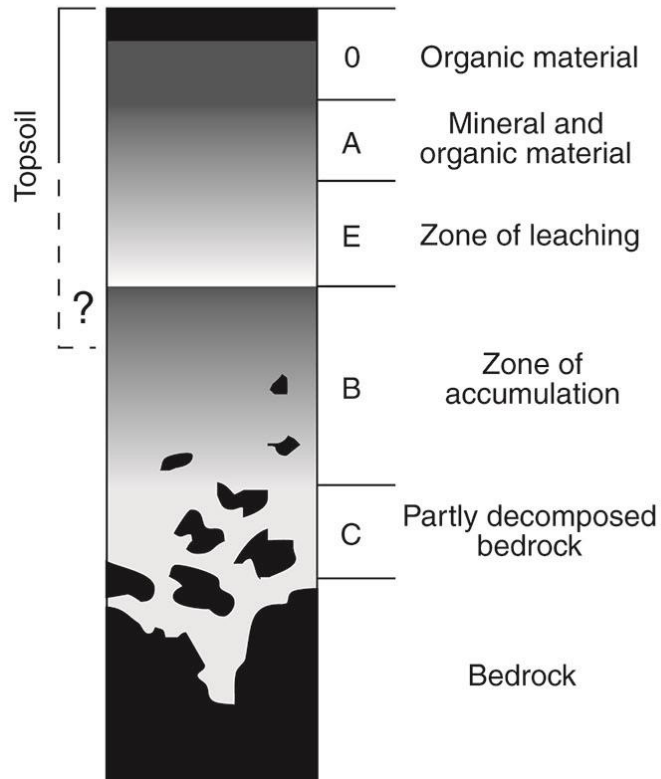


Talc mining in Afghanistan

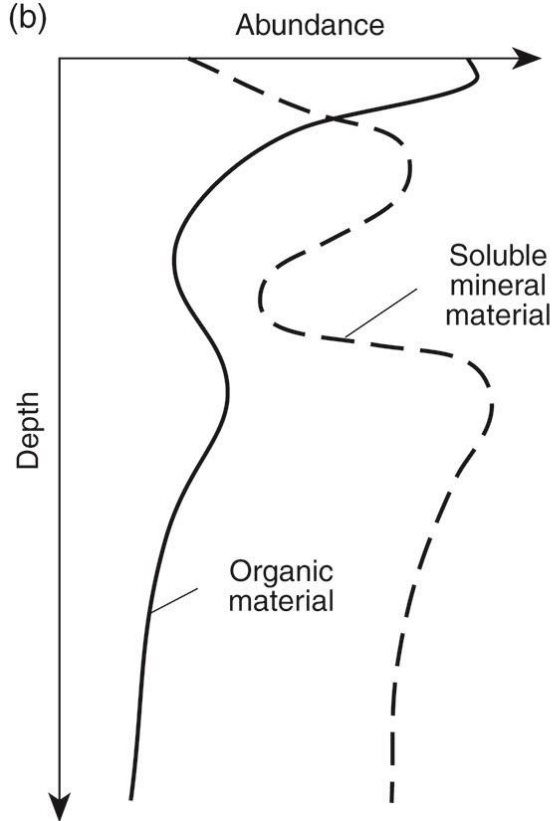


Leaching of the soil (concentrating minerals)

(a) Typical soil profile



(b)



(c)



Laterites (soil enriched in Fe, Ni, Al)



Some vocabulary:

- Talc
- Laterite
- Evaporites (salts)
- Playas