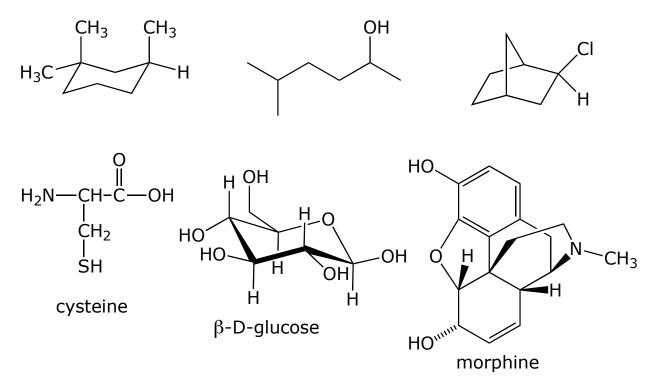
## WORKSHOP 7a

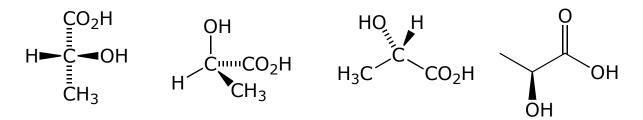
## **Stereochemistry**

1. Identify all the stereocenters in the following compounds.

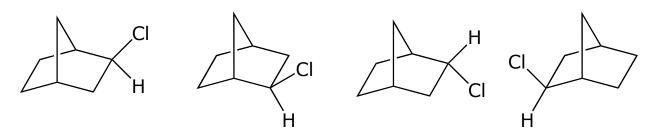


2. a) Determine the relationship between the compounds in each of the following pairs.b) Identify the optically active (chiral) compounds and any **meso** compounds.

c) Assign configuration (R or S) to all stereocenters.



Note: among four structures, there are six pairwise relationships.



3. For each of the following reactions, give a careful representation of the structure of the product and predict whether the product will be optically active, a racemic mixture or achiral. **Explain your choice.** 

a. 
$$(+)-2$$
-chlorobutane  $\begin{array}{c} Br_{2}, hv \\ \hline & \\ \hline & \\ in CCl_{4} \end{array}$  2-bromo-2-chlorobutane

b. 
$$(+)-2$$
-chlorobutane  $\xrightarrow{SO_2Cl_2, hv}$  1,2-dichlorobutane  
Several other products are formed.  
This product is separated by gas  
chromatography and collected for analysis.

c. (+)-2-chlorobutane  $\begin{array}{c} \mathsf{KOH} \\ \hline \mathsf{in EtOH} \end{array} \begin{array}{c} \mathsf{HBr in} \\ \hline \mathsf{ether} \end{array}$ 

$$cis-2$$
-butene  $Cl_2$  in  $CCl_4$ 

d.

e. 
$$(+)-1,3$$
-dimethylcyclopentene   
in CCl4

f. (+)-3-methylcyclopentene 
$$\xrightarrow{PtO_2 H_2}$$